

Ecole thématique du CNRS
Spectrométrie de masse à transformée de
Fourier (FT- ICR et Orbitrap)

Trapped ion mobility mass spectrometry

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4 avril 2018



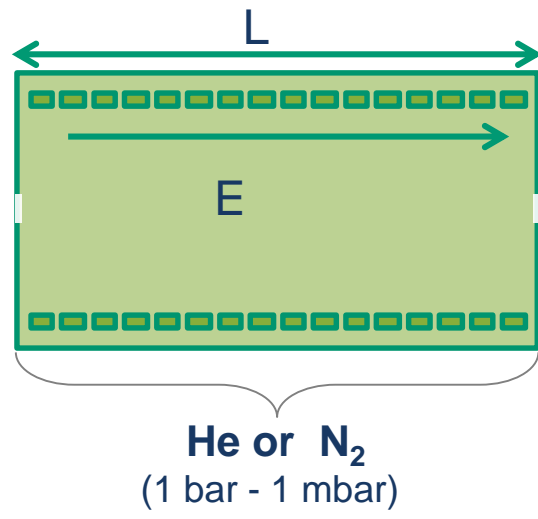
The promises of ion mobility mass spectrometry

- ✓ High speed separation
- ✓ Gas phase separation
- ✓ Identification of ions
- ✓ Measurement of collision cross sections (CCS)
- ✓ Structure elucidation
- ✓ Ion molecule reaction

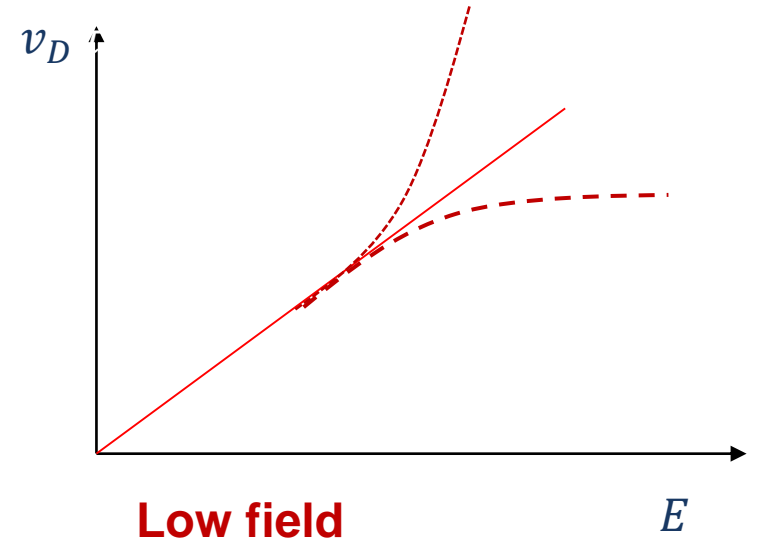
How can it be applied to FTMS ?



Fundamentals of ion mobility



$$v_D = KE$$



Ion mobility K (cm^2/V) Electric field E (V/cm)

Gas number density ($part. m^{-3}$)

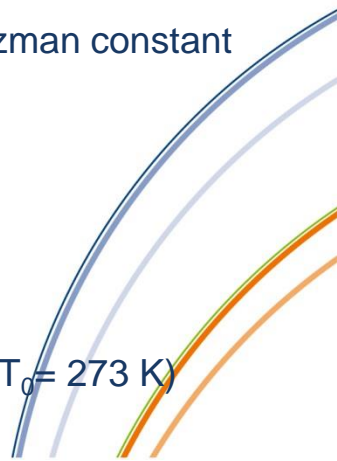
$$N = \frac{n \times N_A}{V} = \frac{P}{kT}$$

Reduced ion mobility K_0

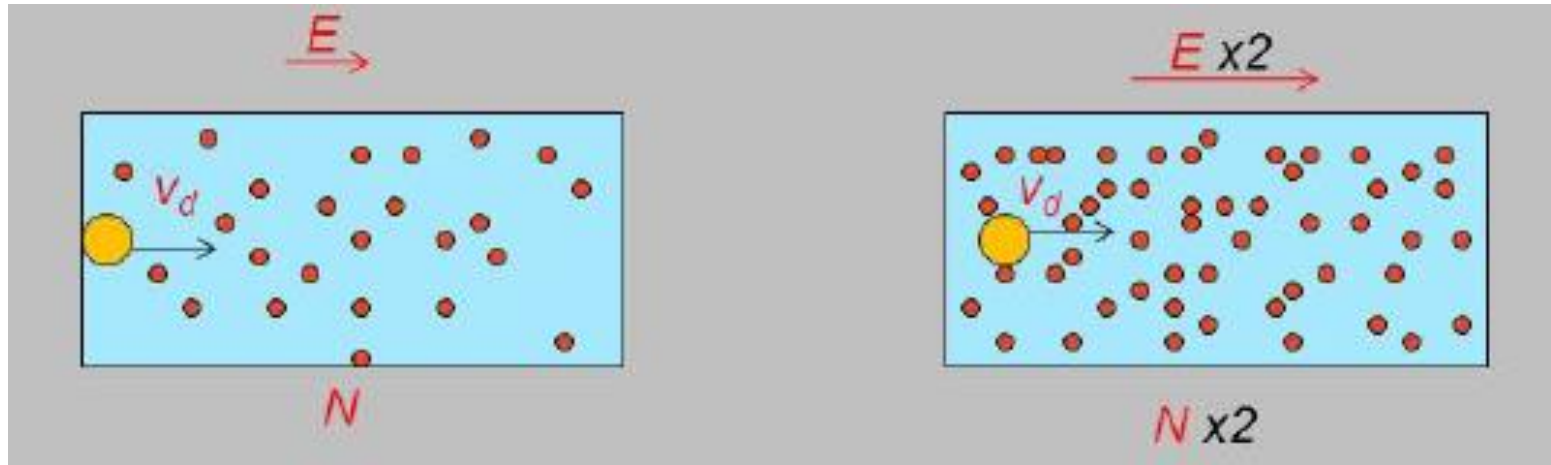
$$K_0 = K \frac{N}{N_0} = K \frac{T_0 P}{T P_0}$$

$N_0 = 2,69 \cdot 10^{25} m^{-3}$ Normal conditions ($P_0 = 1 \text{ bar}$, $T_0 = 273 \text{ K}$)

- N_A Avogadro number
- $k = 1,38 \cdot 10^{-23} \text{ J} \cdot \text{K}^{-1}$ Boltzman constant
- n number of moles
- V volume
- P pressure
- T temperature (Kelvin)



E/N (reduced field) is a key IMS scaling factor

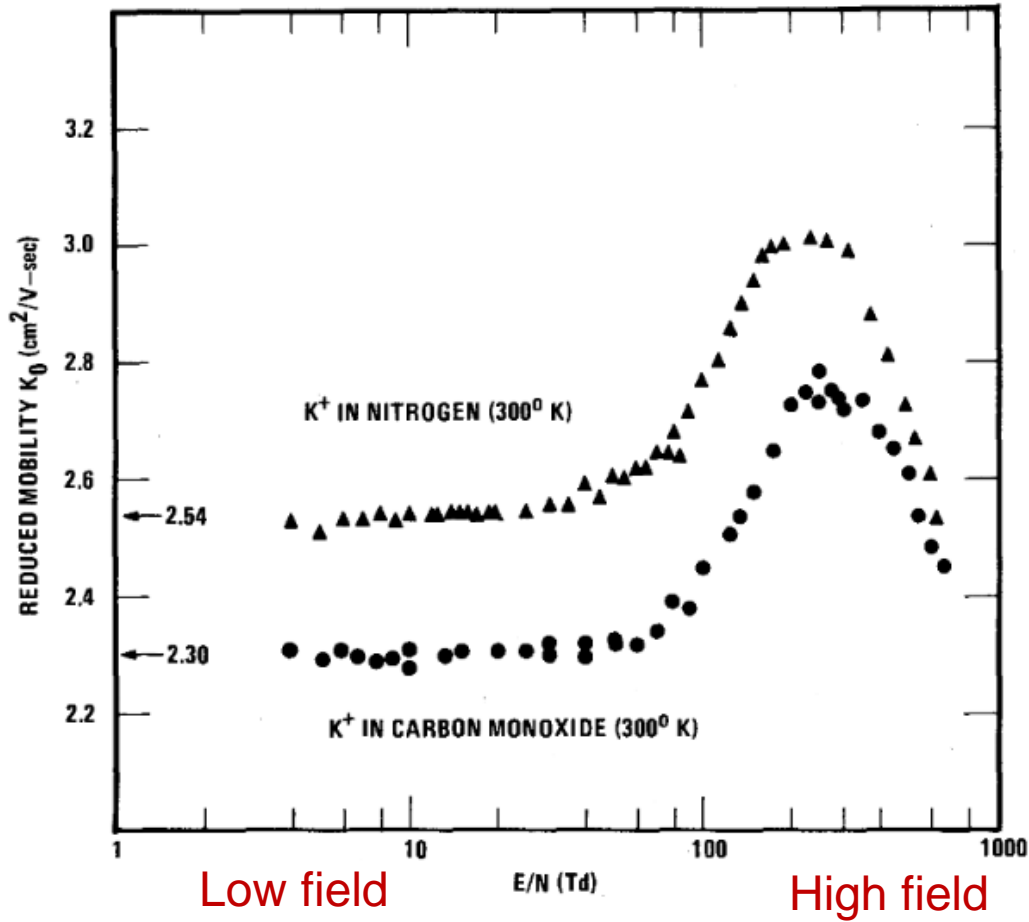


E / N in Townsend 1 Td = 10^{-17} V cm²

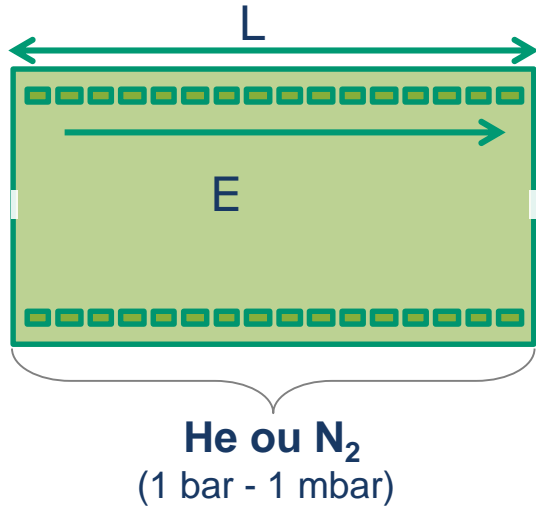
K depends on E/N



E/N (reduced field) is a key IMS scaling factor



Mason – Schamp equation



Low field (E/N) :

Thermal equilibrium between ions and the gas molecules

Ion motion :

Random motion (heat energy) + small drift component (E)

Balance between acceleration (field) and deceleration (collisions)

$$v_D = KE$$

$$K = \frac{\sqrt{18\pi}}{16} \frac{e}{N\sqrt{kT}} \frac{z}{\sqrt{\mu}} \frac{1}{\Omega}$$

$$\mu = \frac{mM}{(m + M)}$$

$e = 1,602 \cdot 10^{-19} \text{ C}$

$k = 1,38 \cdot 10^{-23} \text{ J} \cdot \text{K}^{-1}$ Boltzman constant

T temperature (Kelvin)

N buffer gas number density (m^{-3})

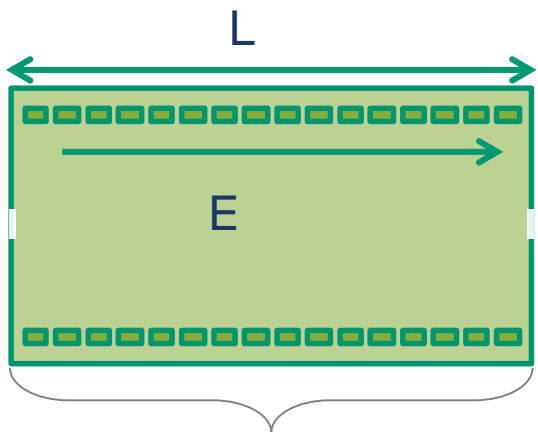
μ reduced mass (molecule and buffer gas) (kg)

z charge number

Ω Collision cross section

Ion mobility measurement

Uniform field drift tube ion mobility



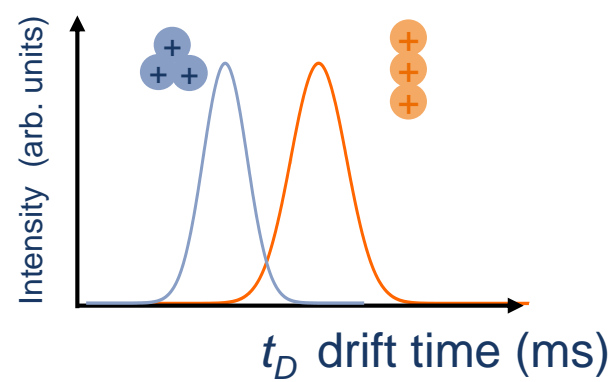
He ou N₂
(1 bar - 1 mbar)

$$v_D = KE$$

Linear function t_D vs $1/V$:

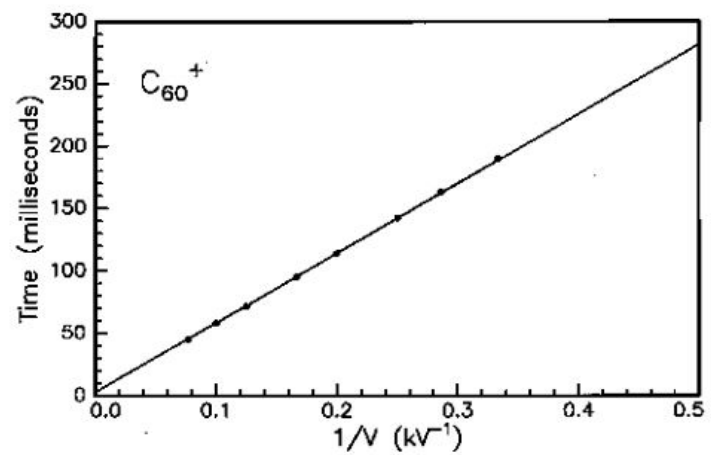
Slope L^2/K

Extracted ion mobility spectrum (m/z)
Arrival time distribution (ATD)



$$t_D = \frac{L}{KE}$$

$$t_D = \frac{L^2}{K} \frac{1}{V}$$



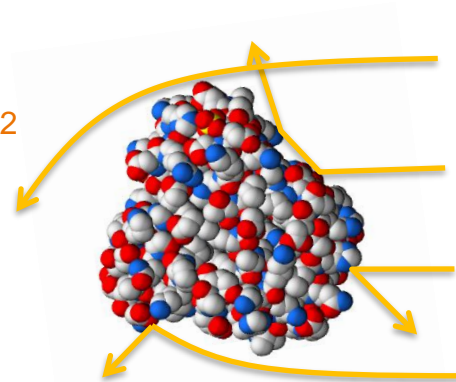
Collision cross section

$$K = \frac{\sqrt{18\pi}}{16} \frac{e}{N\sqrt{kT}} \frac{z}{\sqrt{\mu}} \frac{1}{\Omega}$$

$$t_D = \frac{L}{KE}$$

$$t_D \propto \frac{\sqrt{\mu}}{z} \Omega$$

Collision cross section (CCS) with He or N₂ : Ω_{He} ou Ω_{N_2}
 area Å² accounting for the scattering process
 in the drift tube: ion conformation in the gas phase



$$\Omega_{\text{avg}}^{(1,1)} = \frac{1}{4\pi^2} \int_0^{2\pi} d\theta \int_0^\pi d\varphi \sin \varphi \int_0^{2\pi} d\gamma \frac{\pi}{8}$$

average over all ion rotations

$$\times \left(\frac{\mu}{k_B T} \right)^3 \int_0^\infty dg e^{-\mu g^2 / 2k_B T} g^5$$

average over all impact velocities

$$\times \int_0^\infty db 2b(1 - \cos \chi(\theta, \varphi, \gamma, g, b))$$

momentum transfer

Types of ion mobility mass spectrometry

Low field (E/N)

Time dispersive

Bath gas (fixed)

Uniform field drift tube (DTIMS)

Agilent, TOFWerk

Travelling wave (TWIMS)

Waters

Mobility selective

Gas flow – ion confinement

Trapped ion mobility (TIMS)

Bruker

Gas flow – spatial dispersion

Differential mobility analyzer (DMA)

www.seadm.com

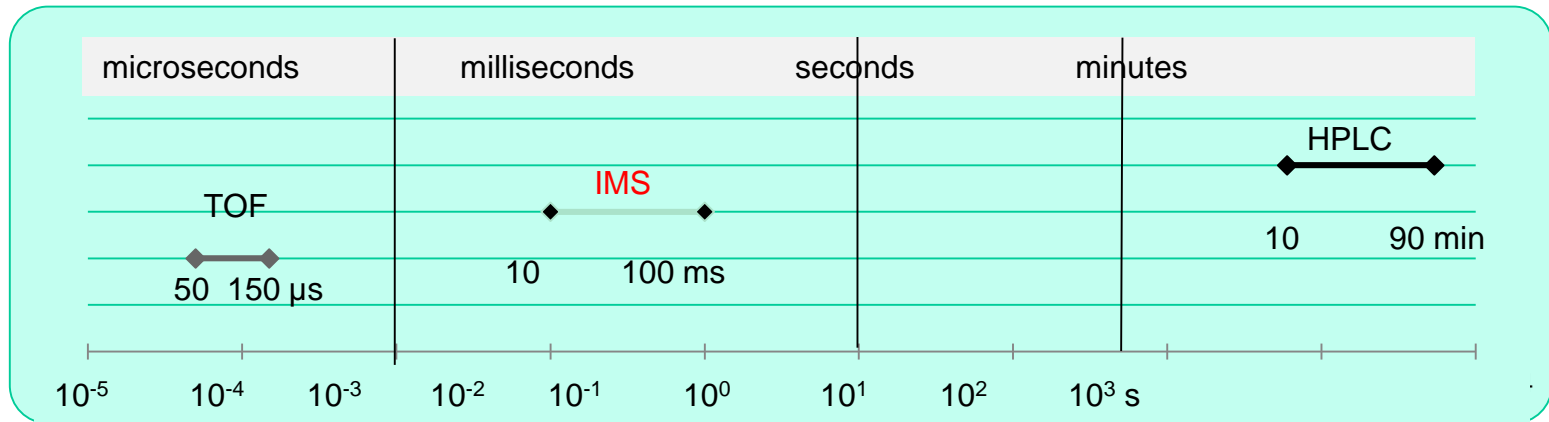
High field (E/N)

Field Asymmetric Waveform
(FAIMS)

Differential mobility
(DMS or DIMS)



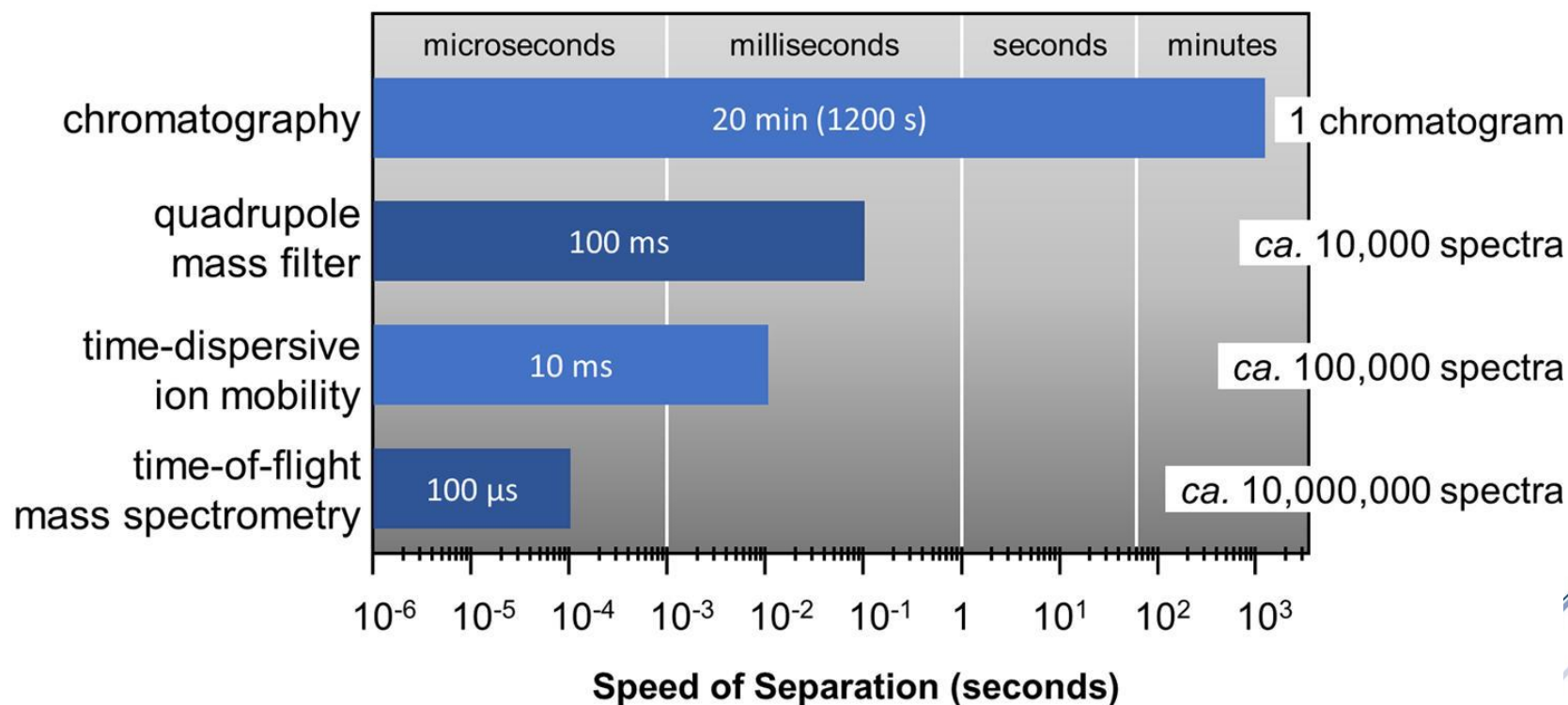
Time ranges of analytical techniques



Time-dispersive ion mobility :
Not compatible with slower mass analyzers

Time ranges of analytical techniques

Nesting of Analytical Timescales



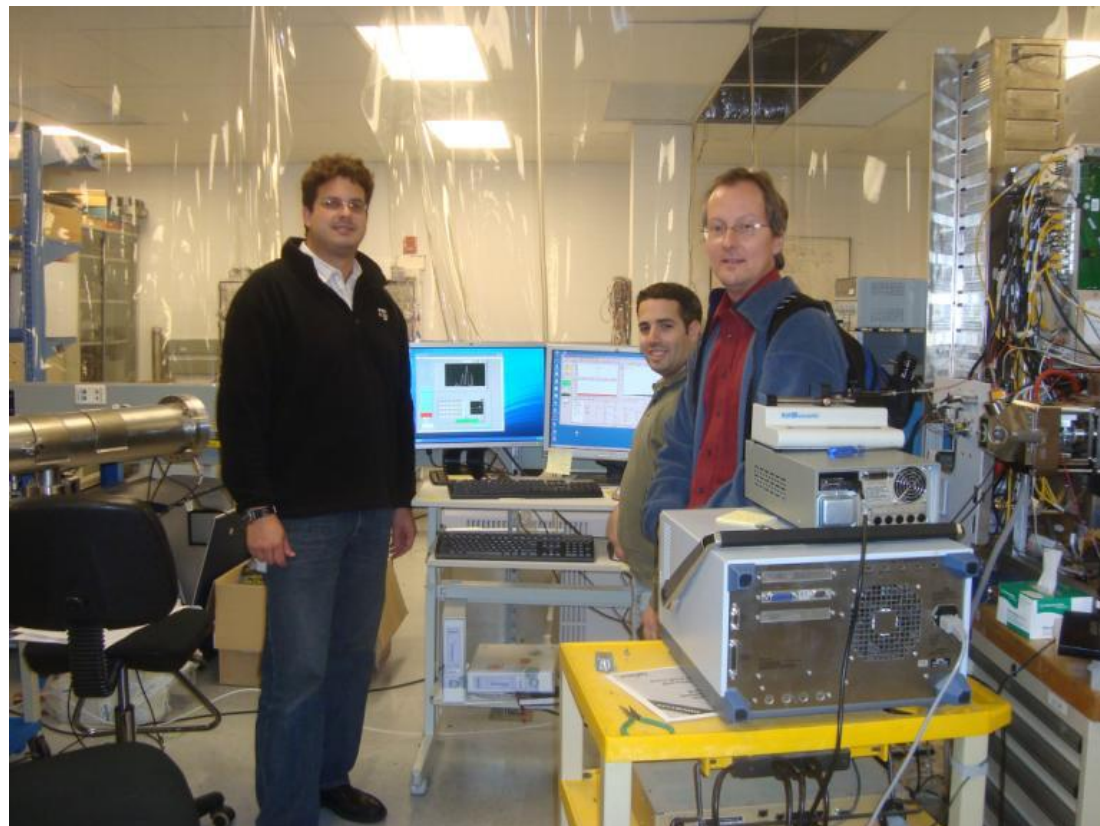
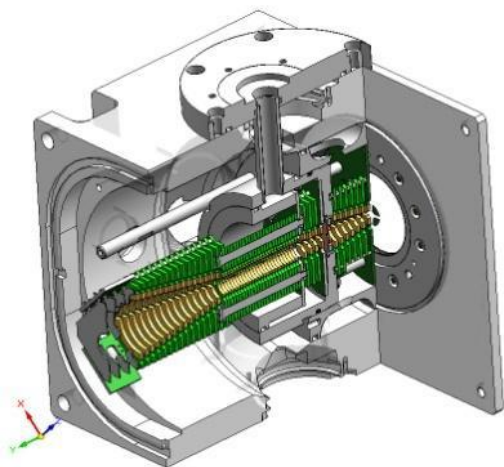
J. C. May et al. *Analytical Chemistry* **2015**, 87, 1422.

Trapped ion mobility spectrometry

2010-2012

Development of the first TIMS prototype using a micro-q-TOF at Bruker Daltonics Inc., Billerica site

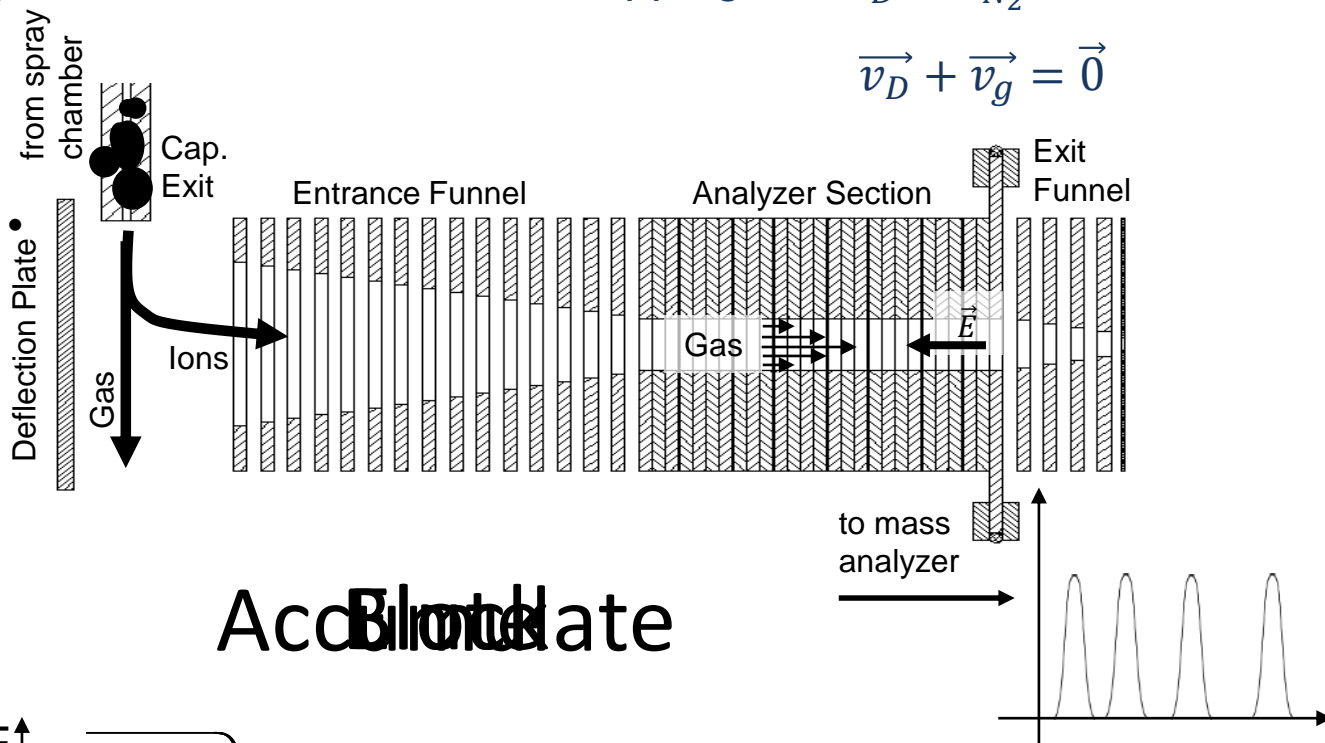
Francisco Fernandez-Lima
Melvin Park
Mark Ridgeway



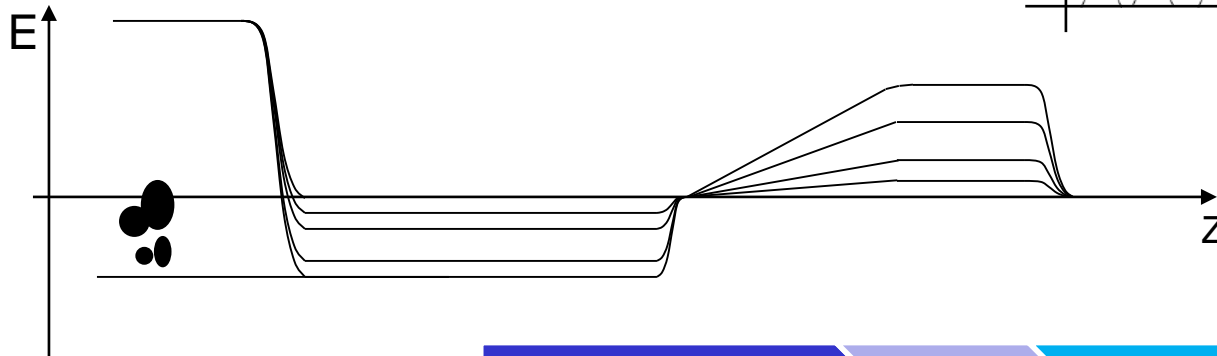
Trapped ion mobility spectrometry

TIMS

Trapping: $\vec{v}_D + \vec{v}_{N_2} = \vec{0}$
 $\vec{v}_D + \vec{v}_g = \vec{0}$



Accumulate



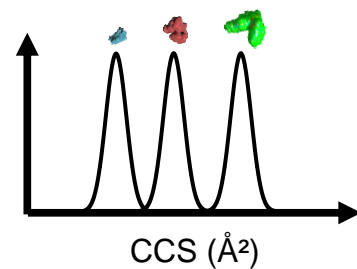
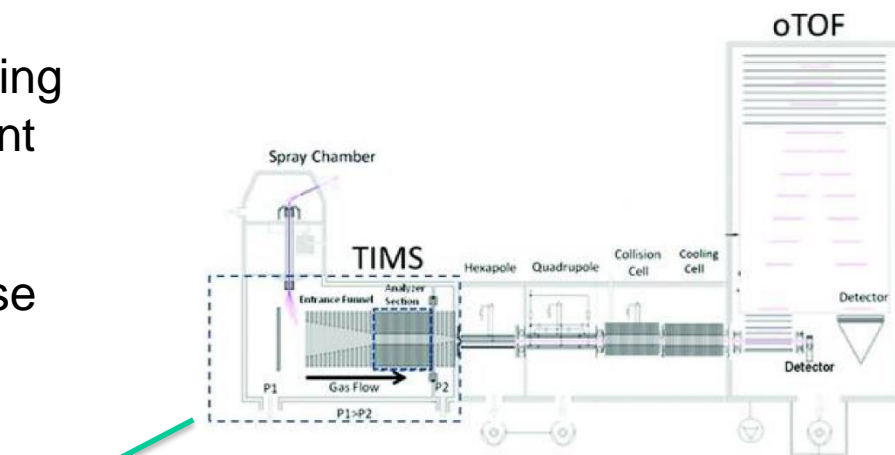
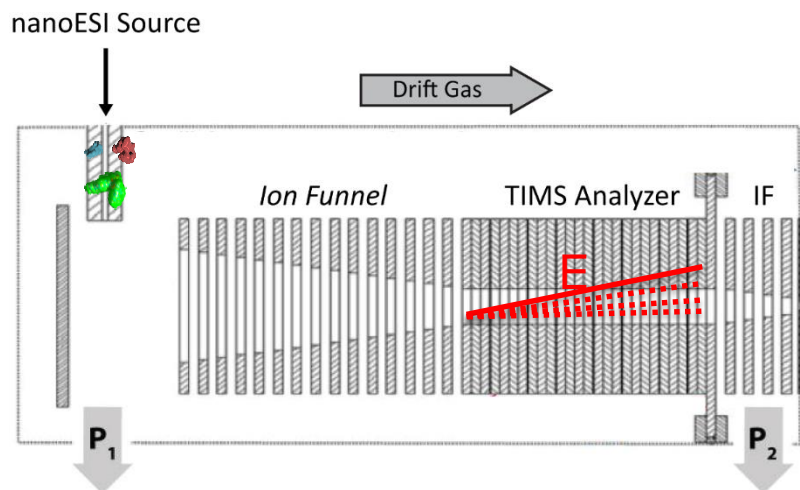
Accumulate

Trap

Elute

Trapped ion mobility spectrometry

- Holding the ions stationary in a flowing buffer gas by an electric field gradient
- Collision Cross Section (CCS): conformation of ions in the gas phase



Unique to TIMS: higher CCS elute first.

Adjustable ion mobility resolution

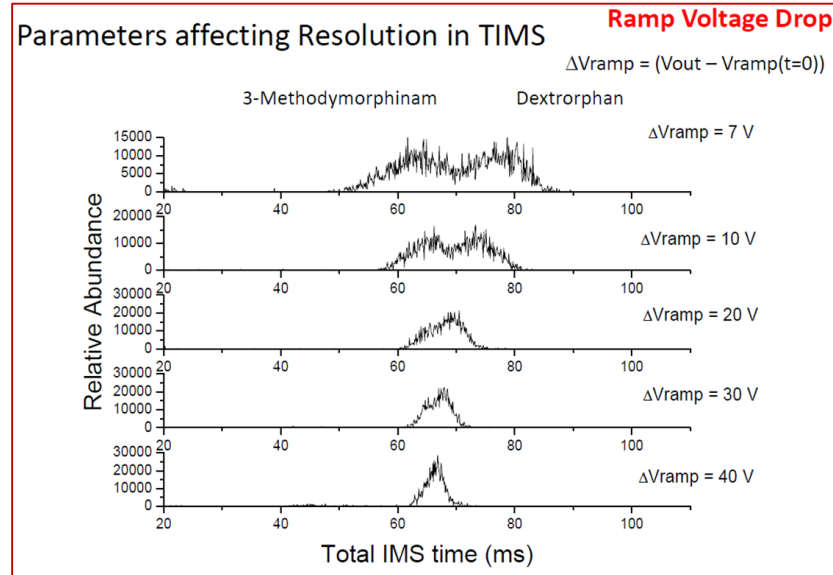
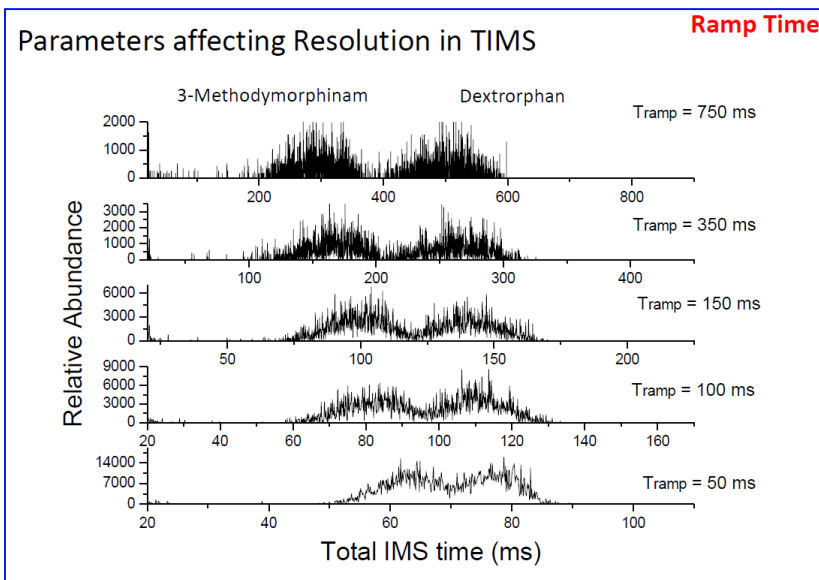
Resolution in drift tube ion mobility

$$R = \frac{t_d}{\Delta t_{FWHM}} = \sqrt{\frac{zeLE}{16 \ln 2 kT}}$$

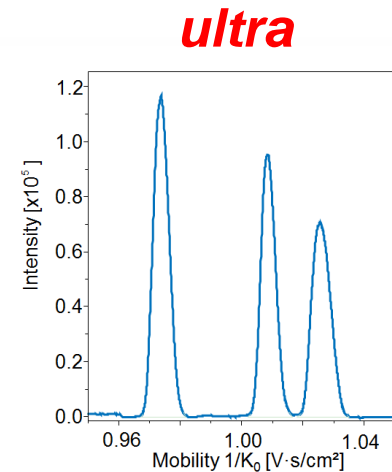
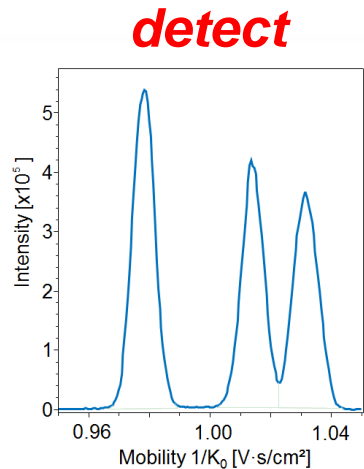
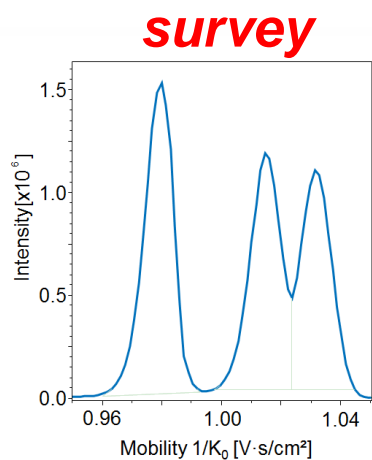
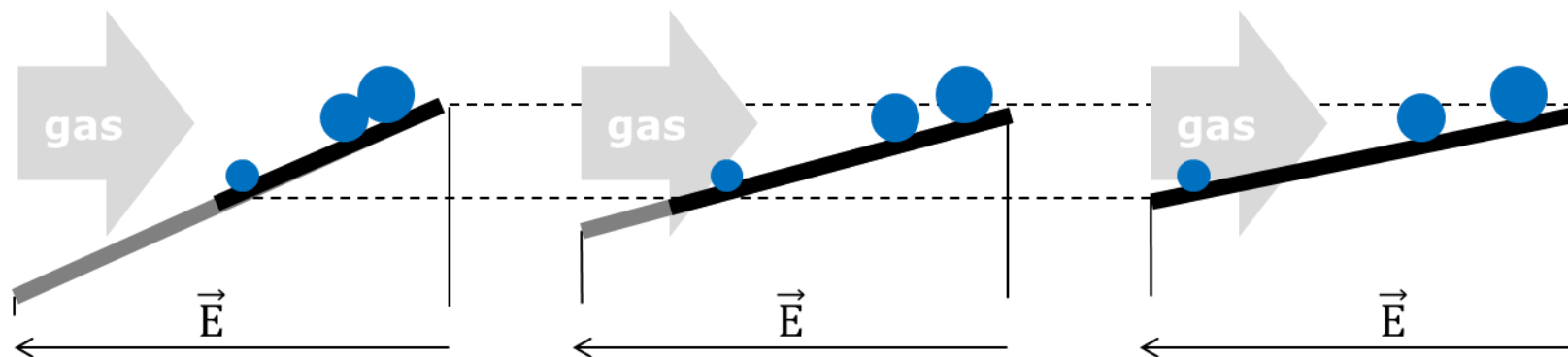
Resolution in TIMS

- Scan rate ($V.ms^{-1}$)
- Velocity of gas (pressure difference)
- TIMS trap time
- TIMS ramp voltage

$$R = \sqrt{\frac{q v_{gt_p} E_e}{16 \ln 2 k_b T}}$$

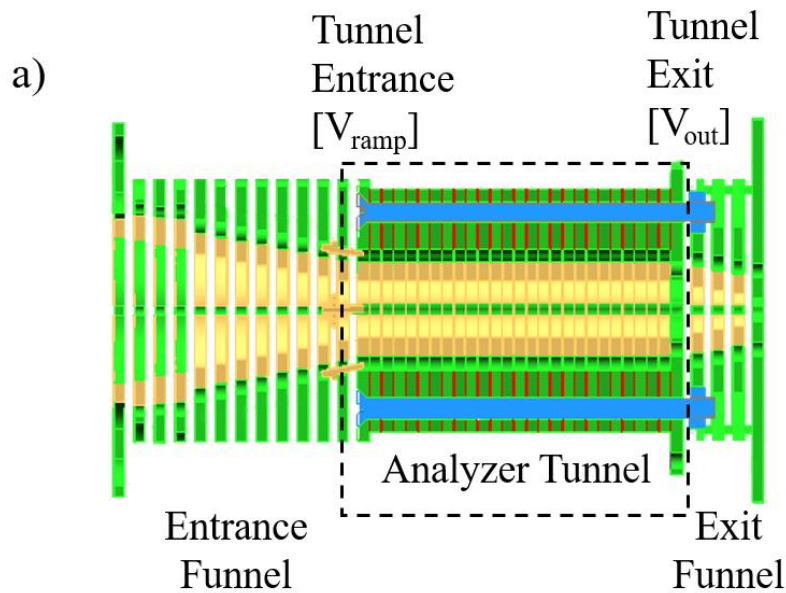


Adjustable ion mobility resolution

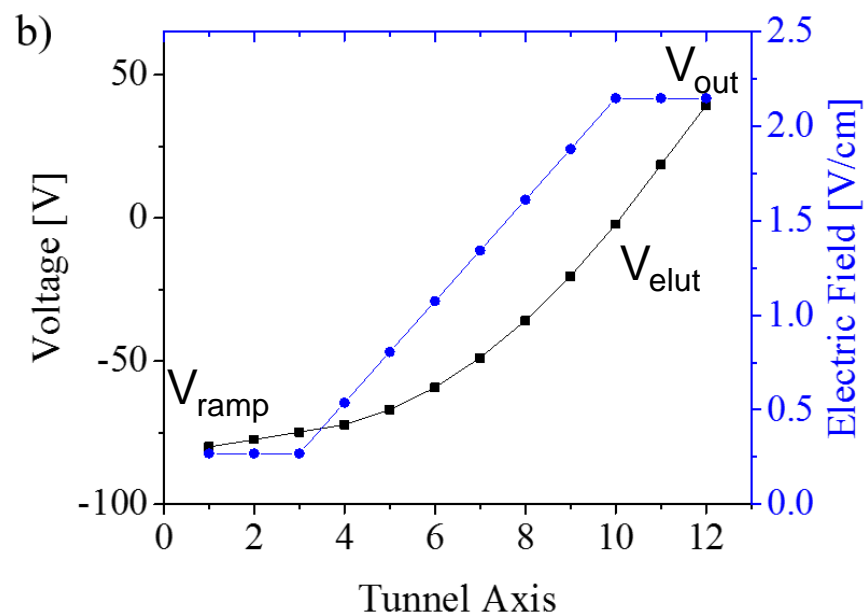


► **imeX™ Technology: *patent pending***

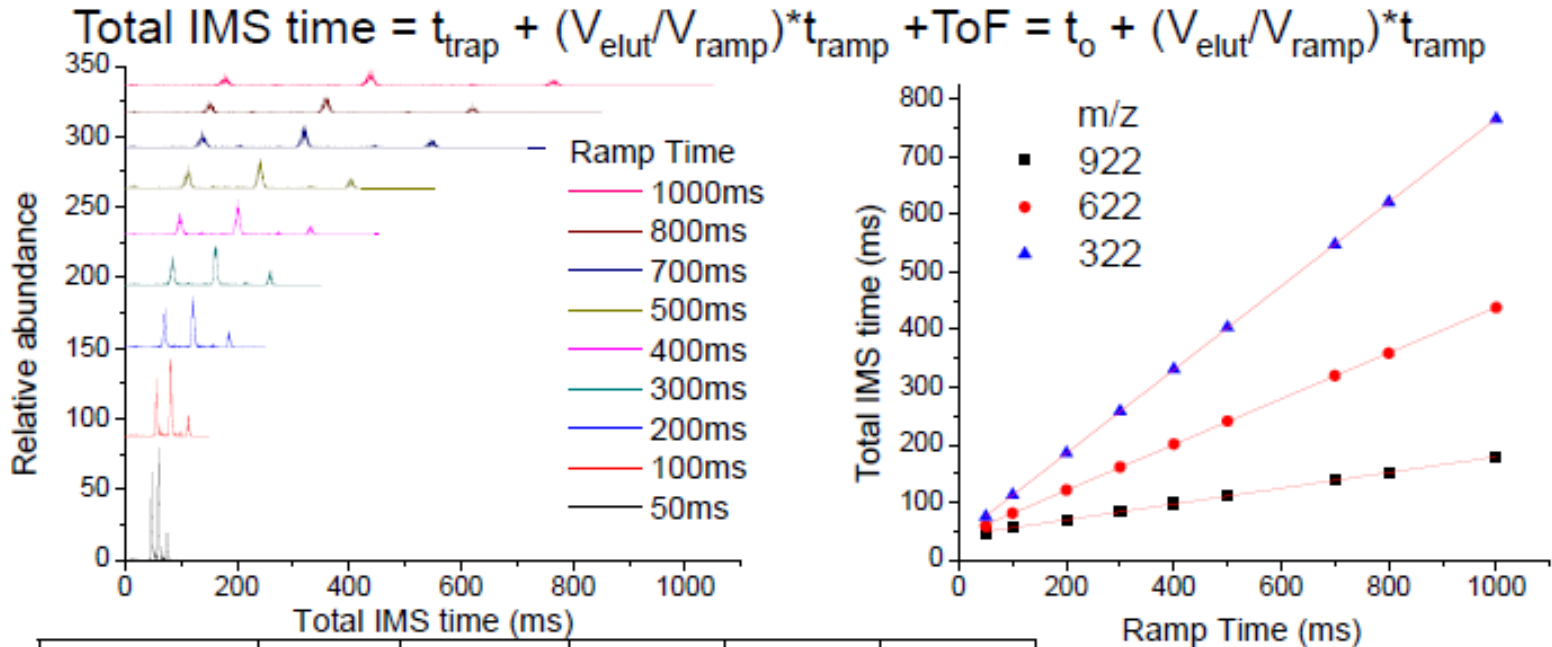
Trapped ion mobility spectrometry



Voltage and electric field



Ion mobility K_0 and CCS measurement

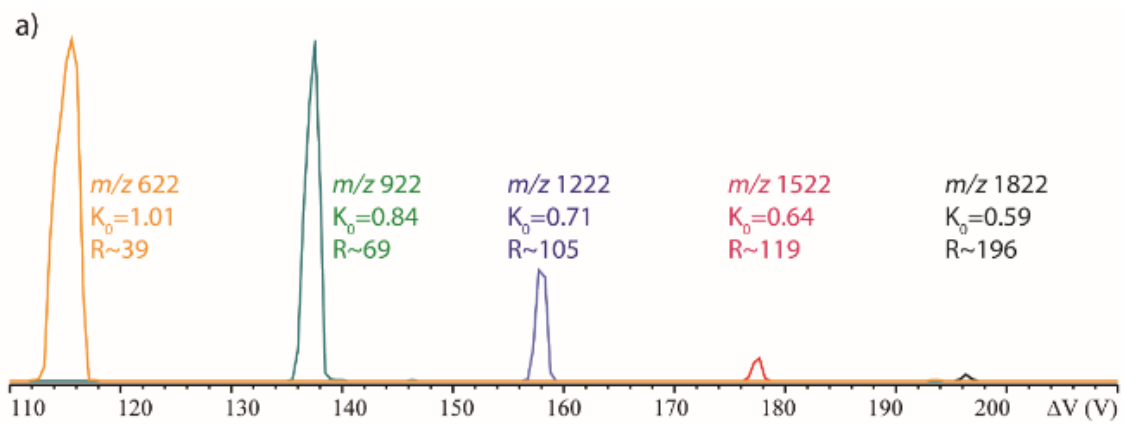


Molecular ion (m/z)	To (ms)	$V_{\text{elut}}/V_{\text{ramp}}$	V_{elut} (V)	K_0 (cm^2/Vs)	CCS (\AA^2)
Tunemix (322)	40.21	0.725	43.50	1.3760	151
Tunemix (622)	41.66	0.397	23.82	1.0130	202
Tunemix (922)	42.63	0.136	8.16	0.8350	243

$$K_x = v_g / E_x$$

$$K_x = A * (1/(V_{\text{out}} - V_{\text{elut}}(x)))$$

Ion mobility K_0 and CCS measurement

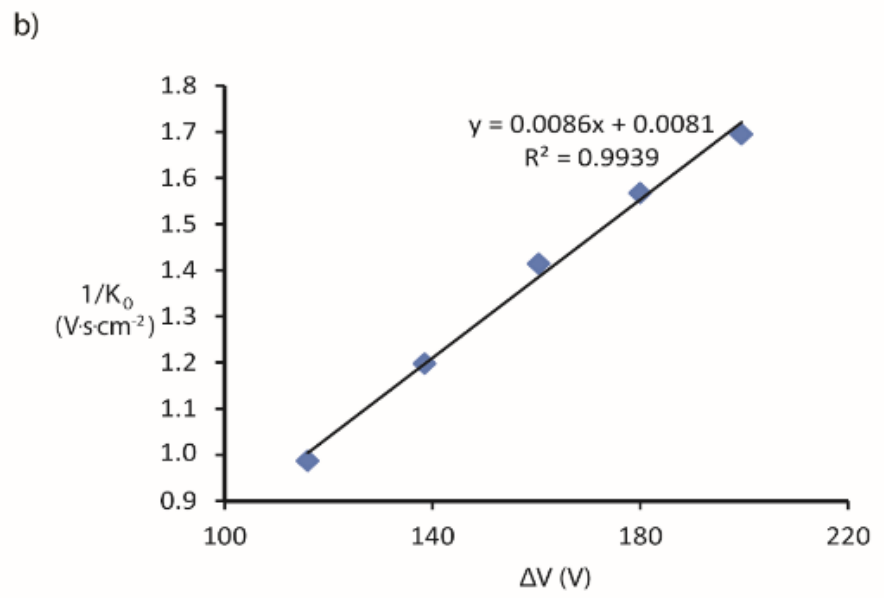


$$KE_e = v_g = \text{constant}$$

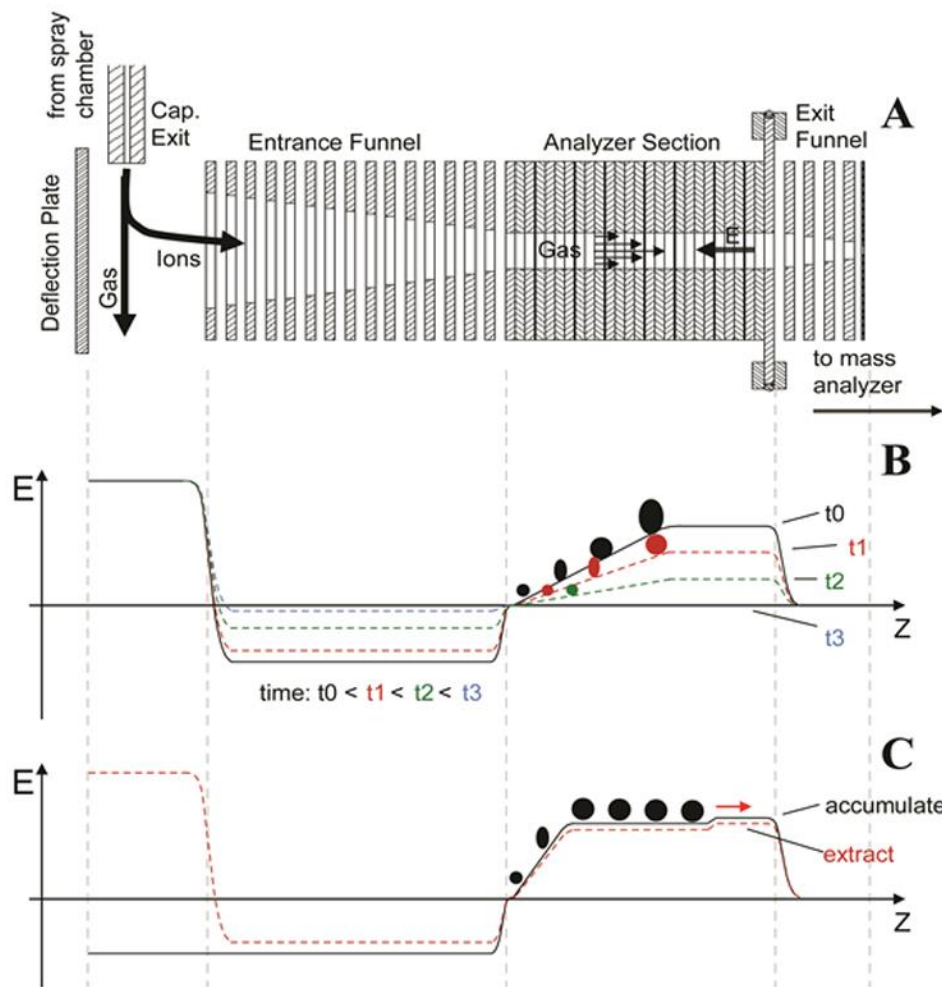
$$K_0 = K \frac{N}{N_0} = K \frac{T_0}{T} \frac{P}{P_0}$$

$$E_e = \frac{\Delta V}{\Delta z}$$

$$\frac{1}{K_0} \propto \Delta V$$



Trapped ion mobility spectrometry : adding FTMS



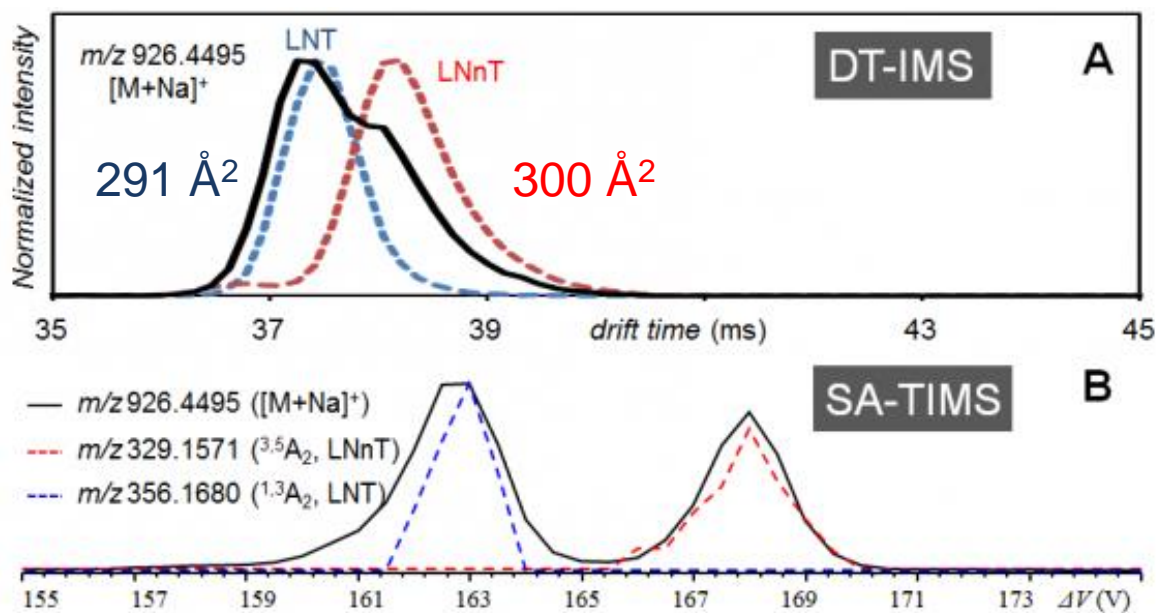
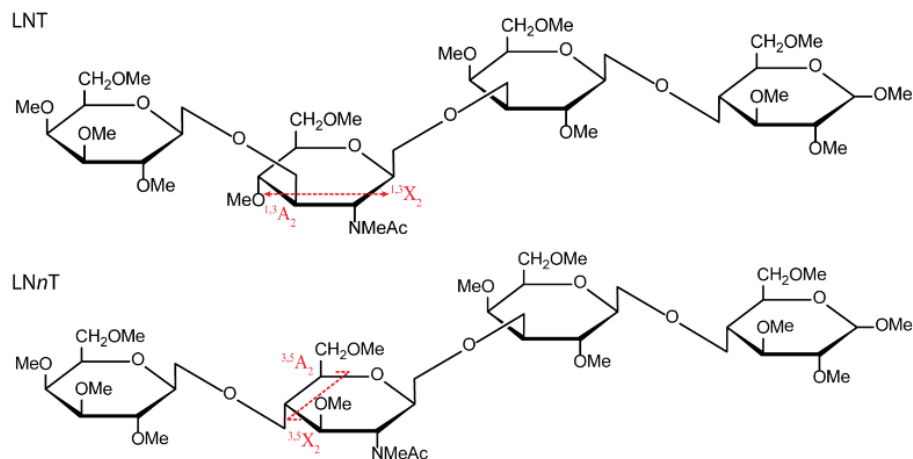
Selected accumulation trapped ion mobility spectrometry

and Gated TMS

Long trapping time :

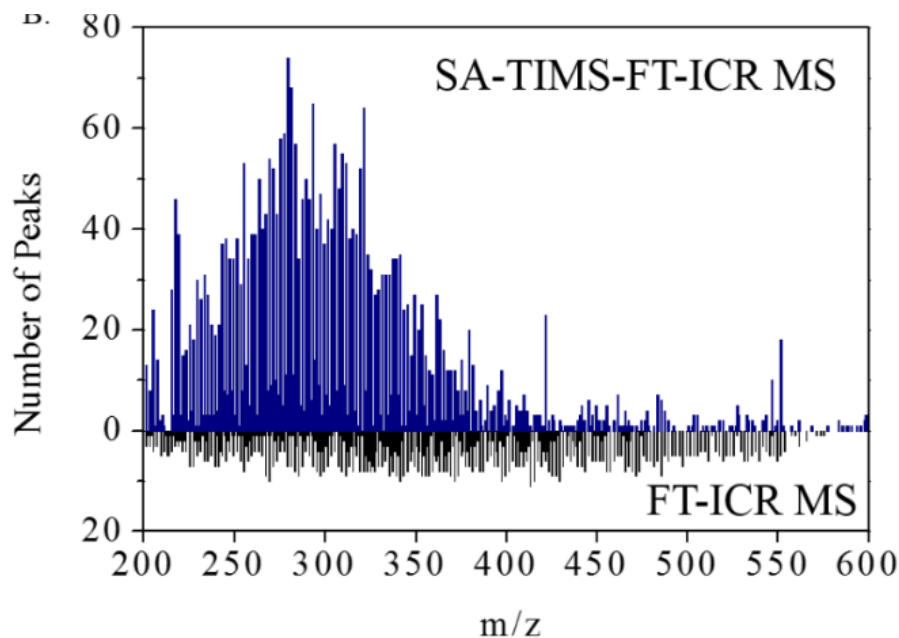
- ✓ higher S/N ratio per scan
- ✓ coupling to slower analysis methods, including FTICR

Trapped ion mobility spectrometry : adding FTMS

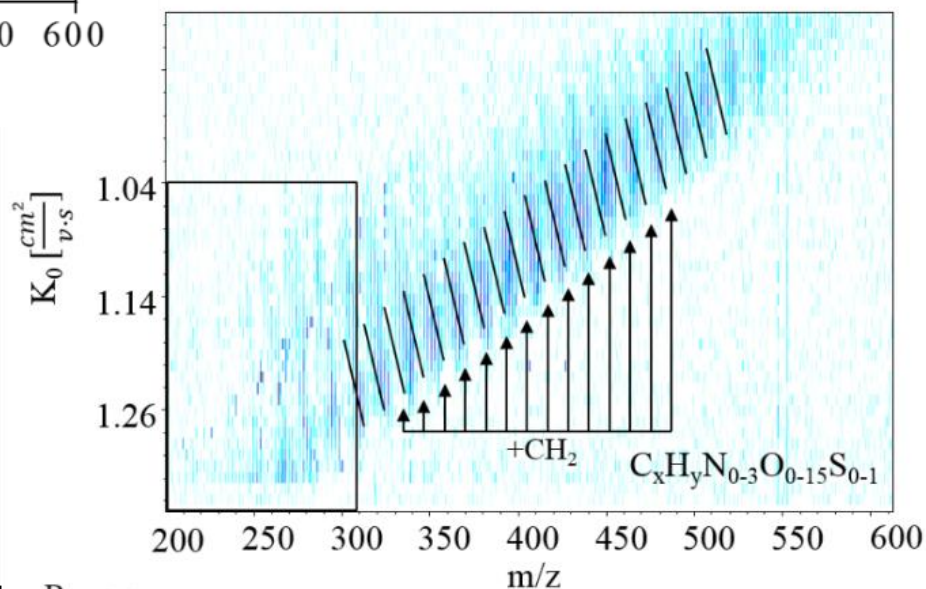


Trapped ion mobility spectrometry : adding FTMS

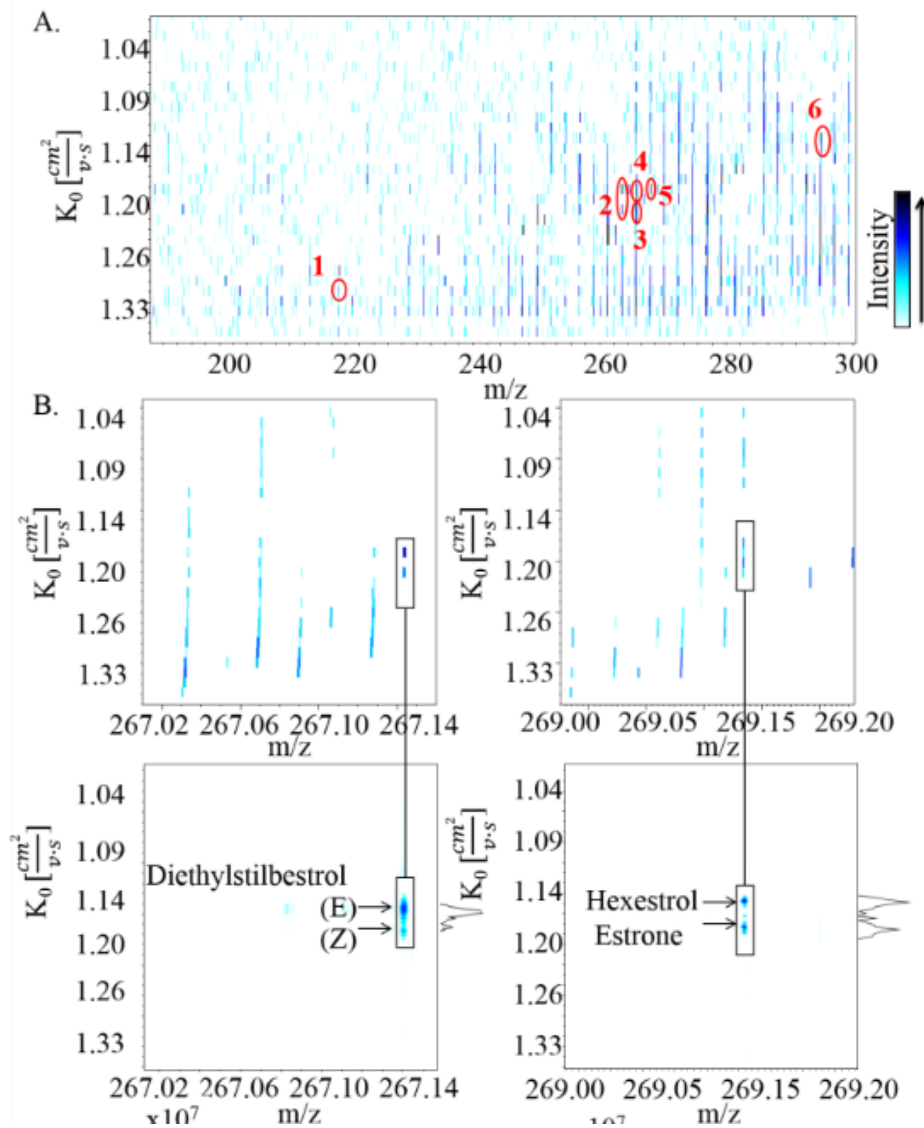
SA-TIMS



- Suwannee River Fulvic Acid Standard II (SRFA) 20 $\mu\text{g/mL}$ doped 5 ppb of
- (1) bisphenol A
 - (2) diethylstilbestrol (Z and E)
 - (3) estrone
 - (4) hexestrol
 - (5) α -estradiol
 - (6) 17- α -ethynylestradiol

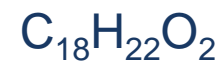
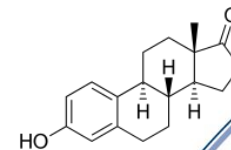
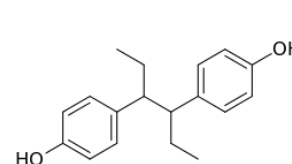


Trapped ion mobility spectrometry : adding FTMS



Suwannee River Fulvic Acid Standard II (SRFA) 20 $\mu\text{g/mL}$ doped 5 ppb of

- (1) bisphenol A
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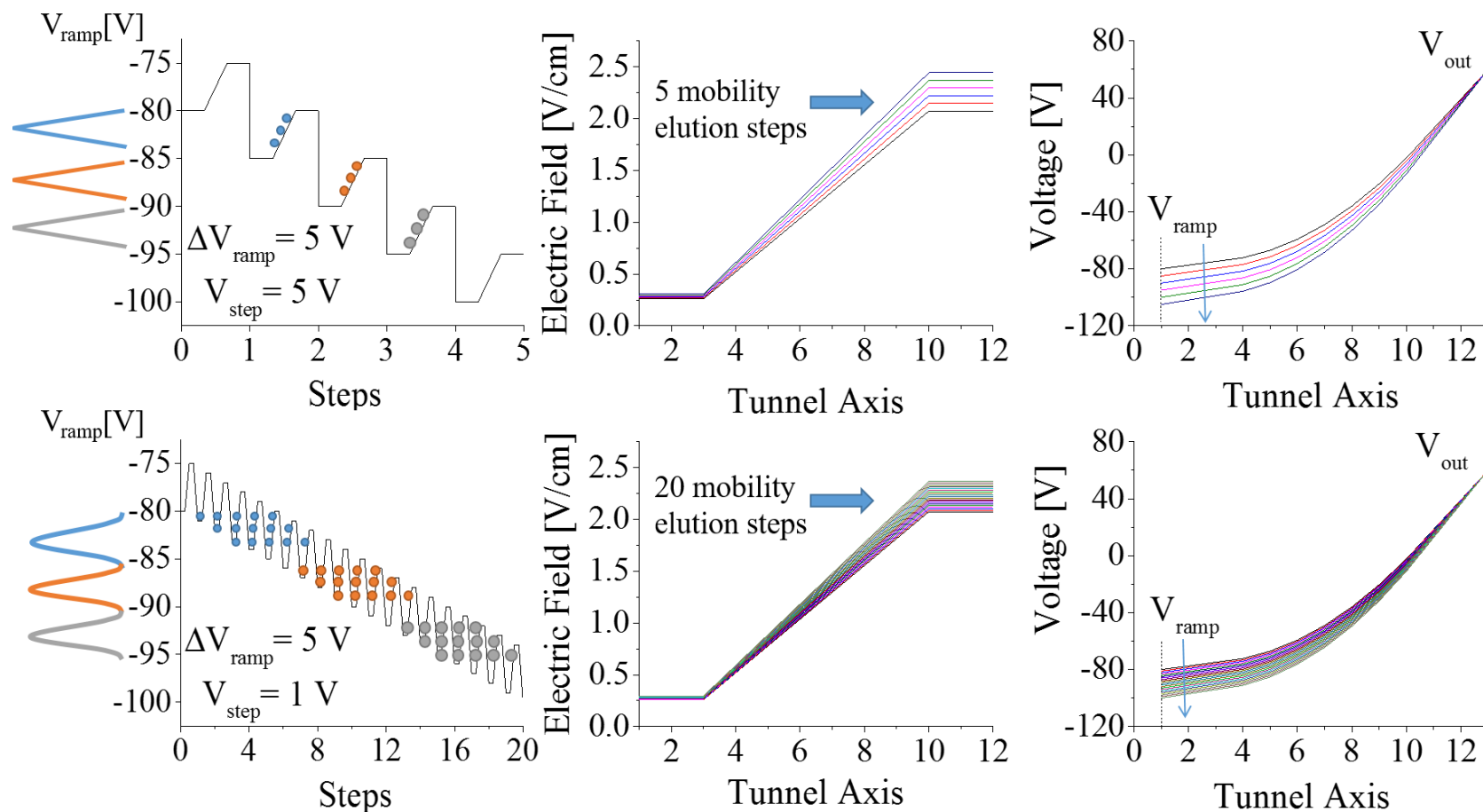


177 \AA^2

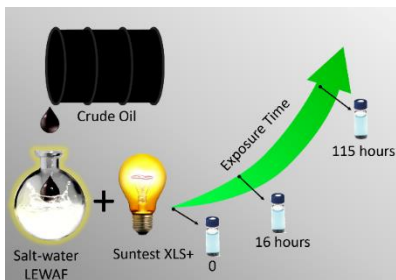
174 \AA^2

Trapped ion mobility spectrometry : adding FTMS

Oversampling SA-TIMS

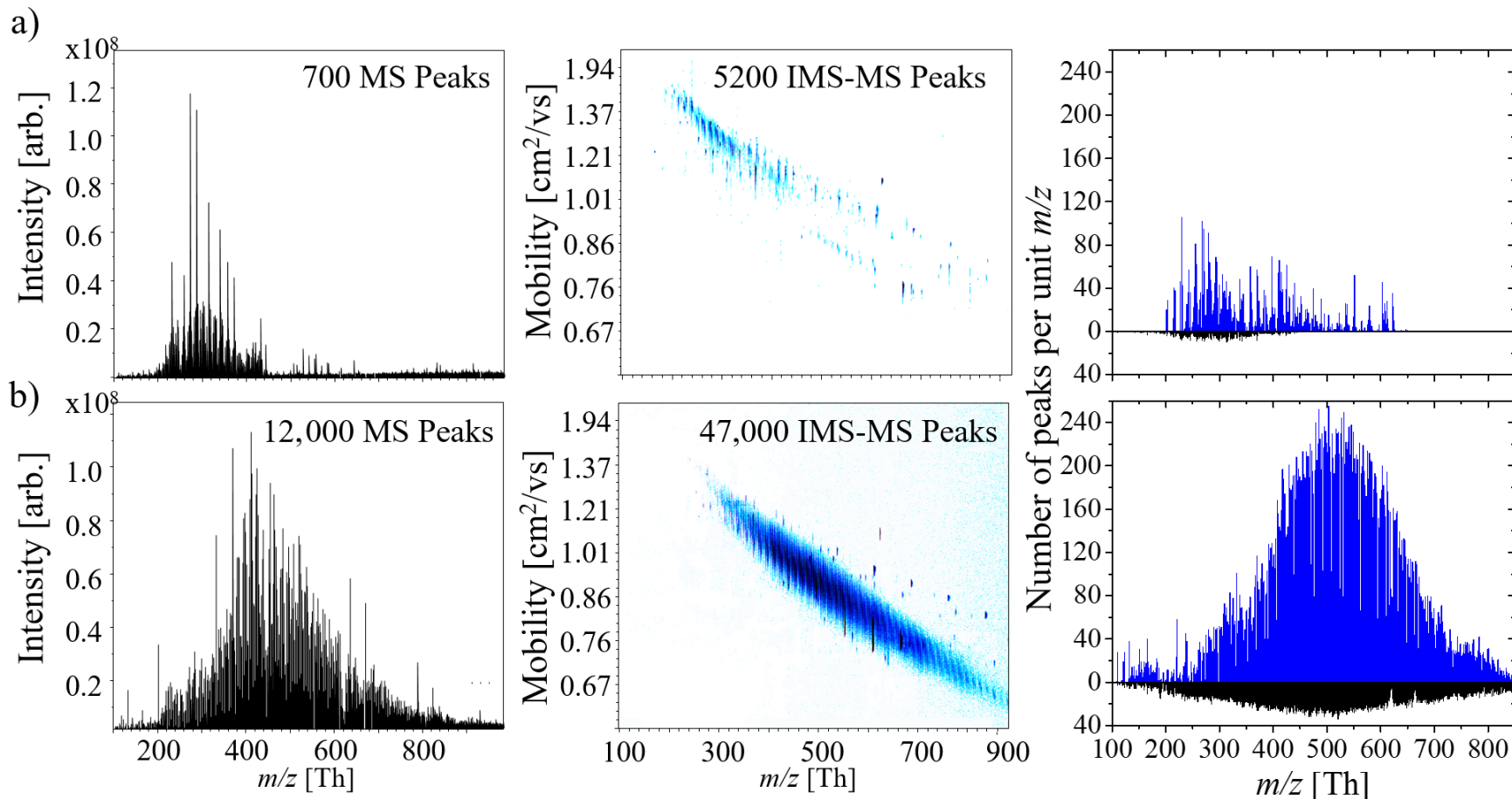


P. Benigni, F. Fernandez-Lima, "Oversampling SA-TIMS coupled to FT-ICR MS: Fundamentals and Applications". Anal .Chem. 2016.



Trapped ion mobility spectrometry : adding FTMS

Oversampling SA-TIMS



P. Benigni, K. Sandoval, C. J. Thompson, M. E. Ridgeway, M. A. Park, P. Gardinali, F. Fernandez-Lima. "Analysis of photo-irradiated water accommodated fractions of crude oils using tandem TIMS and FT-ICR MS" Environ. Sci. Technol. 51 (11) (2017) 5978–5988

Summary

M. E. Ridgeway, M. Lubeck, J. Jordens, M. Mann, M. A. Park.
Trapped ion mobility spectrometry: A short review. *International Journal of Mass Spectrometry* **2018**, 425, 22.

TIMS is a very **flexible** ion mobility technique

Elution of ions : analytical field strength E decreased, E_0 can be measured

Slow/rapid scans : high resolving power / high spectral rates,

Limited voltage range scans = limited mobility range scans

Variety of methods :

original “sequential analysis” TIMS,

selective accumulation TIMS, parallel accumulation TIMS, Gated TIMS,

Oversampling SA-TIMS adapted to FT-ICR

Many future evolutions possible !