

# Spectrométrie de masse à transformée de Fourier (FT-ICR et Orbitrap)

*Ecole thématique du CNRS*

**Cabourg 3-6 avril 2018**

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## LC-FT MS for environmental analysis

Edith NICOL

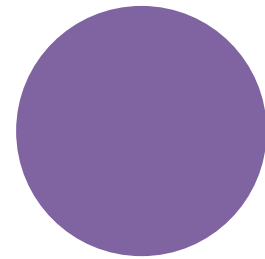
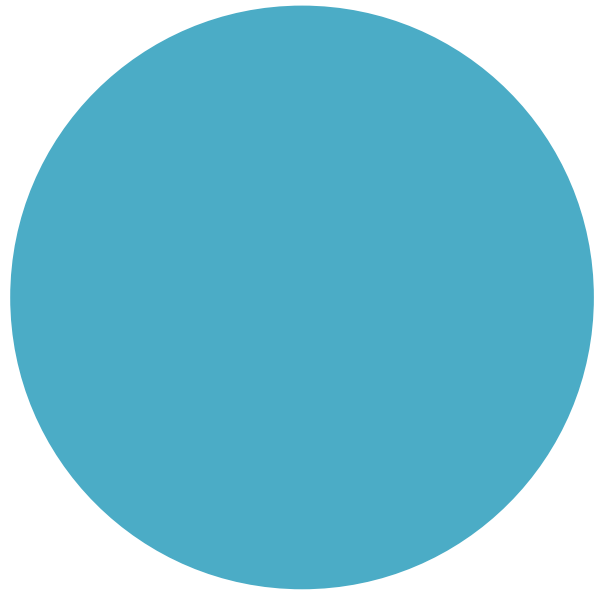
Laboratoire de Chimie Moléculaire

Ecole Polytechnique



# Outline

- I. Pollution
- II. Chromatography
- III. LC-FTMS
- IV. Pollutants analysis



# POLLUTION

Definition, generalities  
and environmental  
impact

# What is pollution?

“Pollution means the introduction by man, directly or indirectly, of substances or energy into the environment, resulting in deleterious effects of such a nature as to endanger human health, harm living resources and ecosystems, and impair or interfere with amenities and other legitimate uses of the environment.”

OECD 1974

# Where does it come from?

Natural: storm, volcanic eruption etc...



Anthropogenic: exhaust gases, industries etc ...



# Types of pollution

Primary

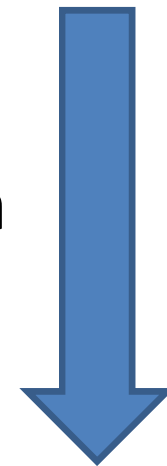
At the root

Secondary

transformation of a pollutant in  
natural environment



degradation



?

# Polluted environments

## Air

Industries, transport, pollen ...



## Soil

Fertilizer, pesticides, waste...



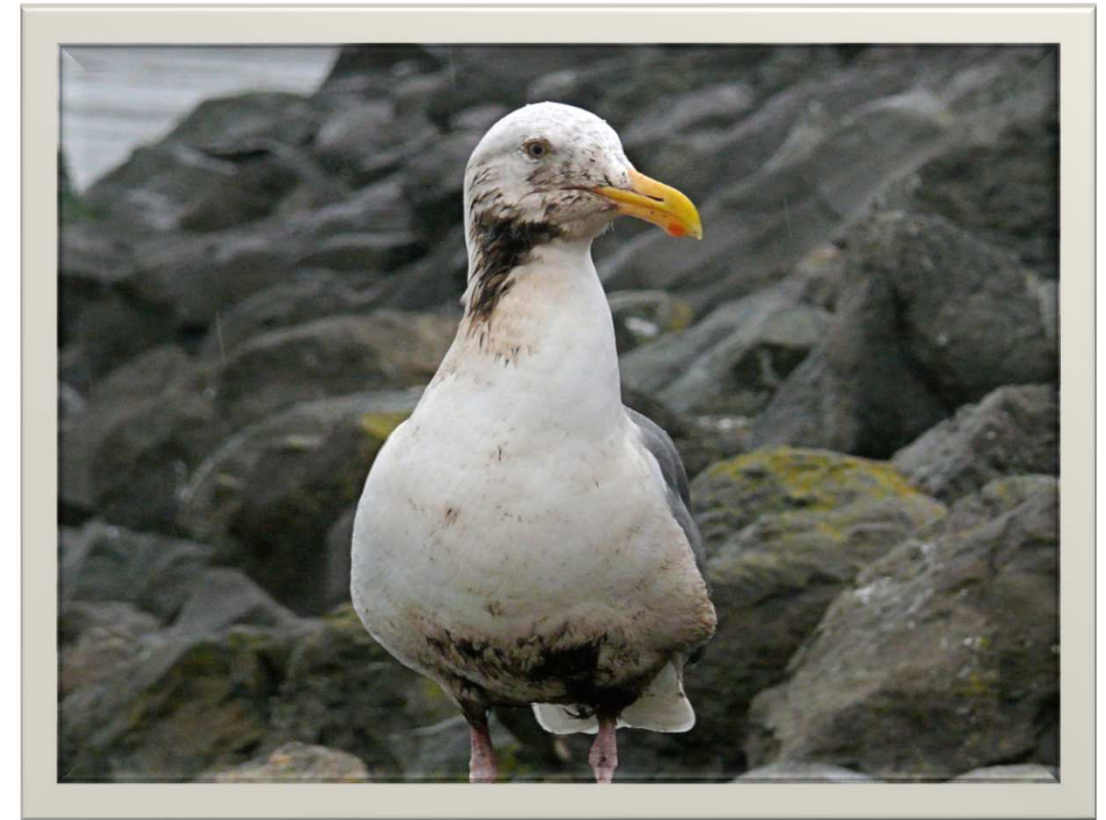
## Water

Fertilizer, industries aqueous waste, pesticides, bacteria...



# Incidence

- One-time
  - oil spill, industrial accident ...
- Persistent (permanent modification of ecological factor)
  - greenhouse gas, pesticides...





# Environmental impact

- Direct or indirect effects on an actor of the environment
- Exposition level
- Ecosystem sensitivity towards the pollutant
- Ecosystem ability to absorb the pollutant



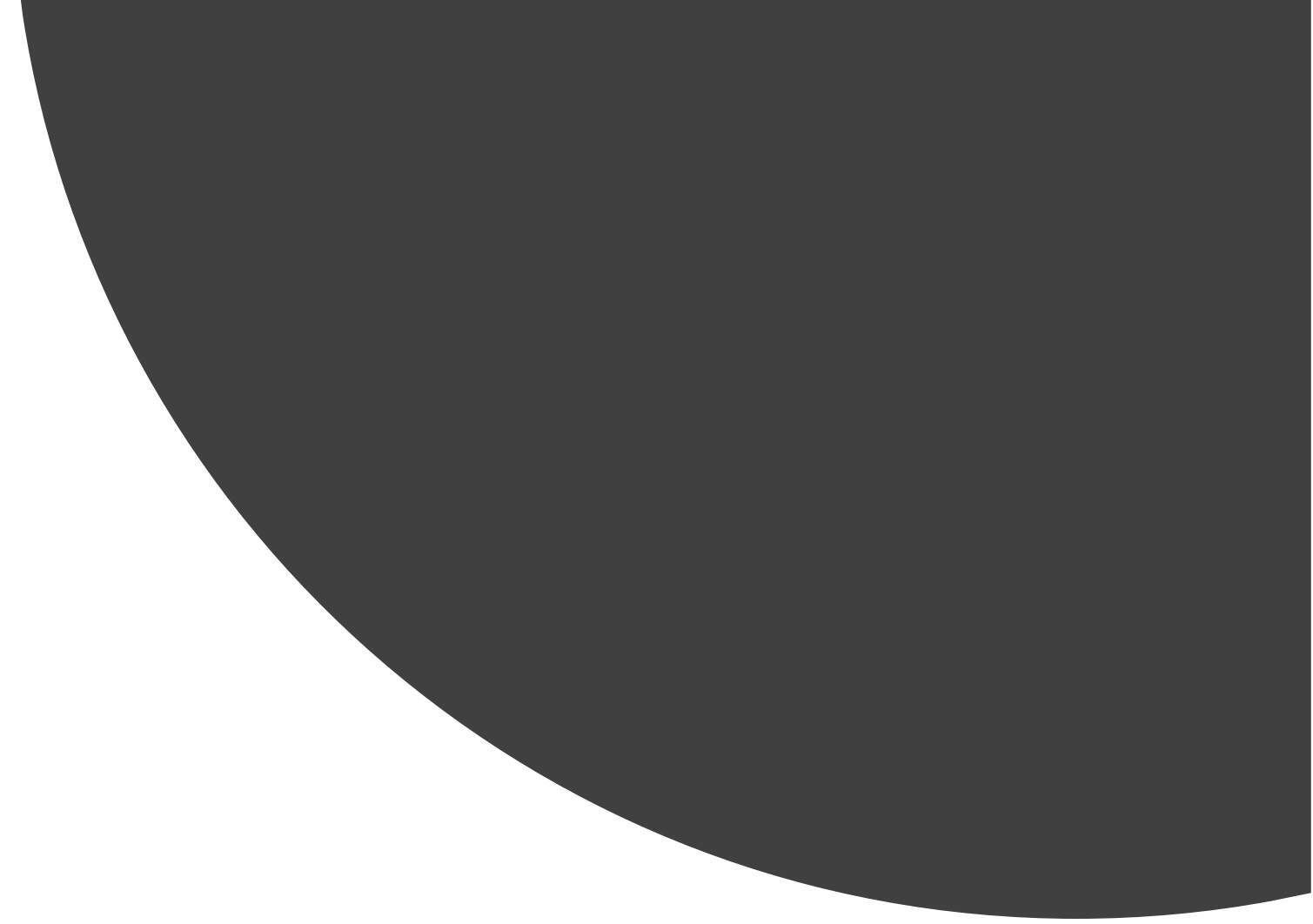
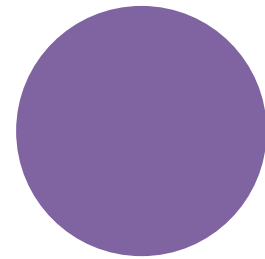
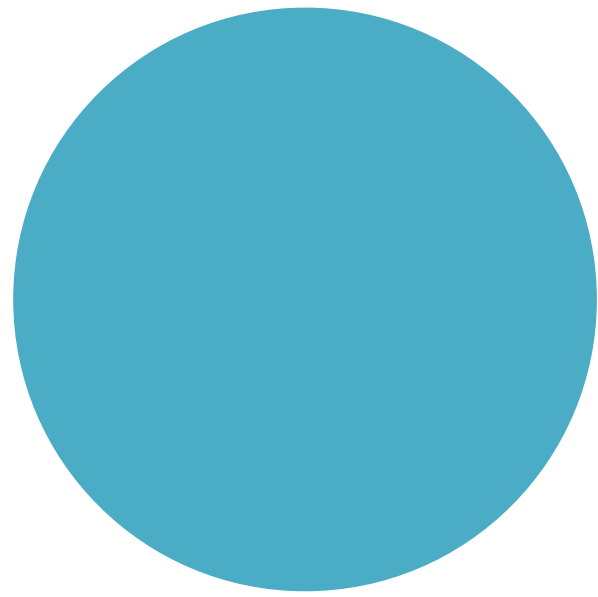
# Problematics

Unknown molecules  
Traces amount  
Big volume  
Complex matrix



→ Sample preparation  
→ Instruments with high sensitivity and selectivity

- Chromatography: separation of compounds
- Mass spectrometry: sensitive and specific, characterization of compounds

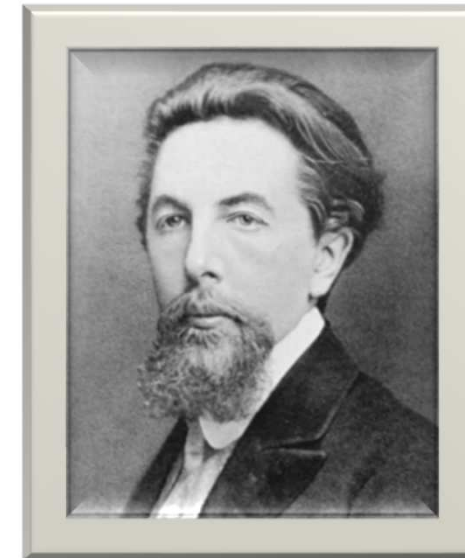
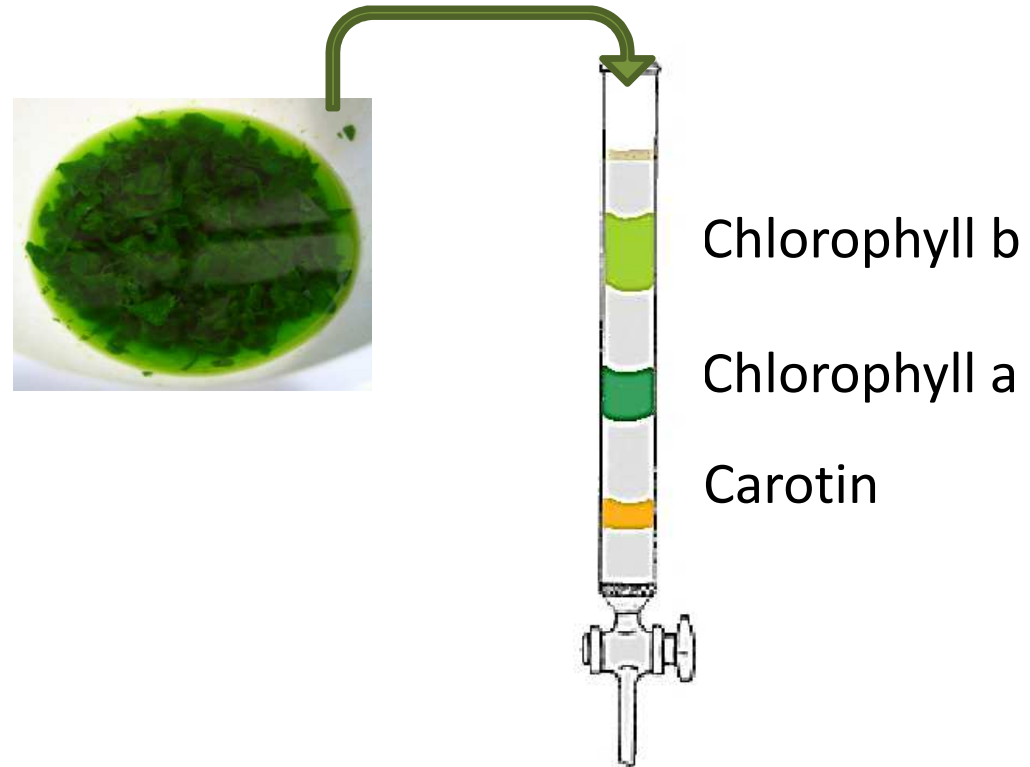


# LIQUID CHROMATOGRAPHY

History, separation,  
resolution

# History

1906 : 1<sup>st</sup> chromatography → plant pigments separation



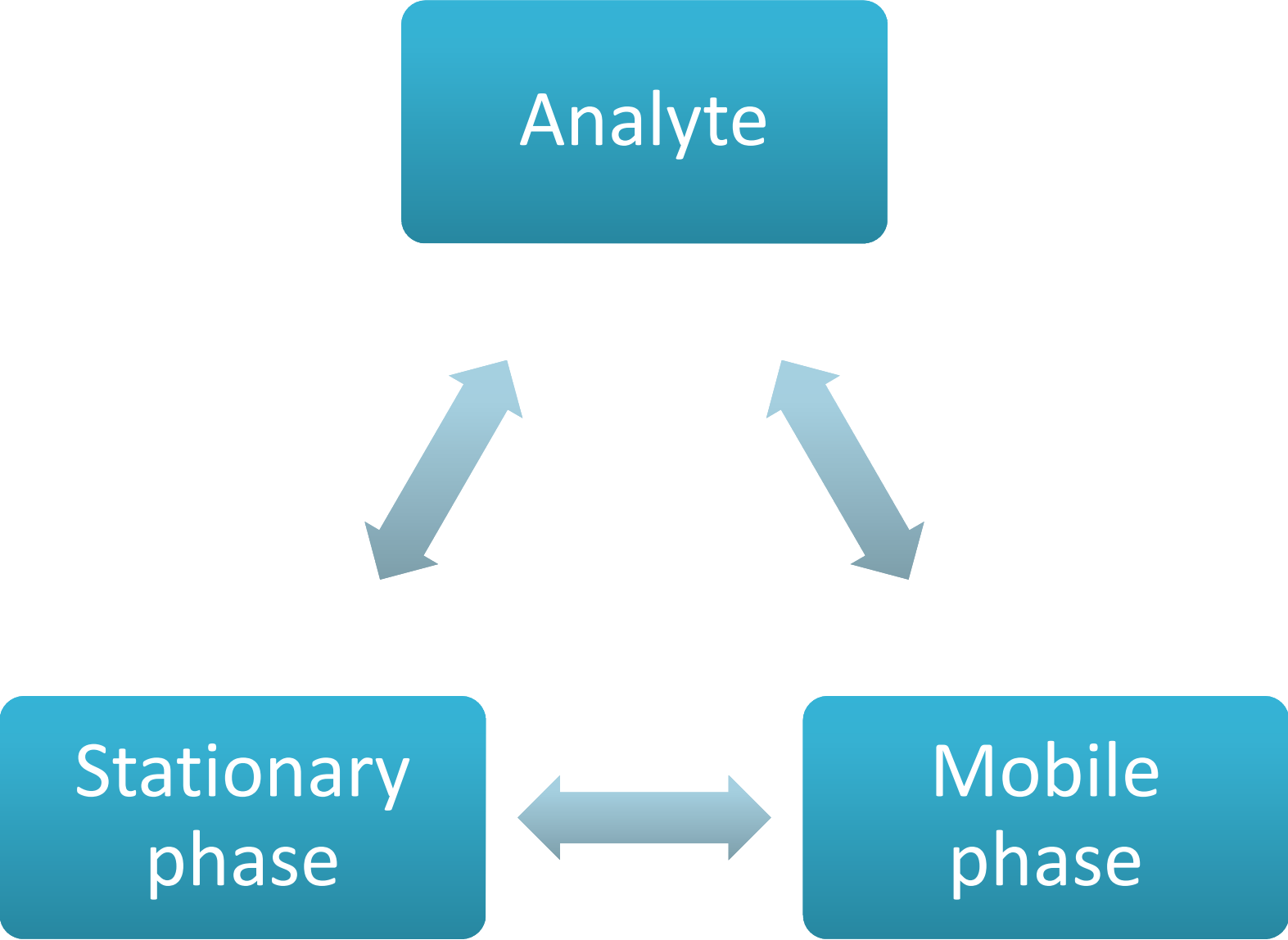
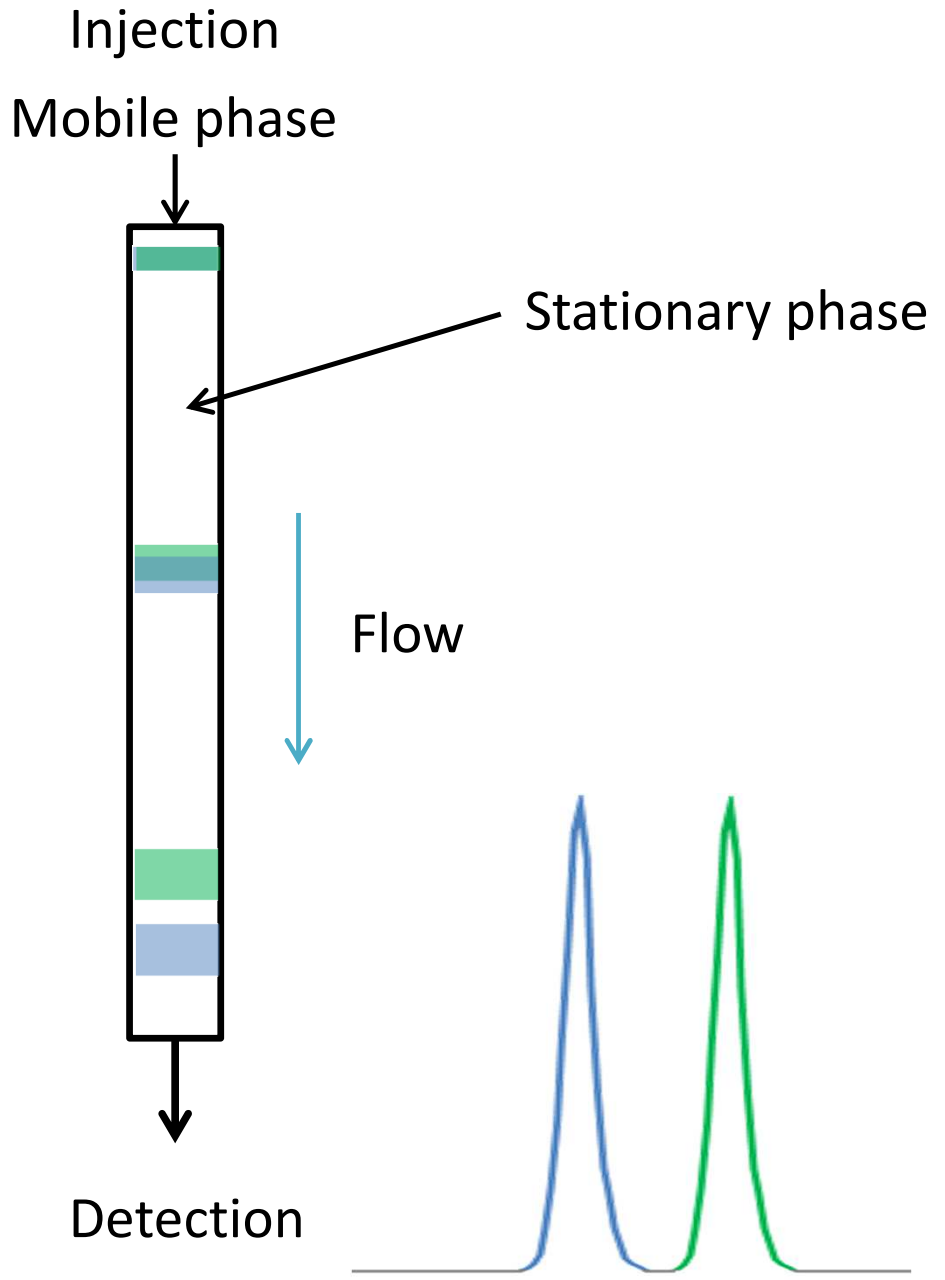
M.S. Tswett, botanist

Chromatography is derived from Greek « *kroma* » (color) and « *graphein* » (to write)

1950 : Gas chromatography

1969 : Liquid chromatography development

# Separation



# Different chromatography

## Gas

### Analyte ?

- Volatile
- Thermally-stable

### Important parameters ?

- Temperature
- Stationary phase composition

## Supercritical fluid

### Analyte ?

- Soluble in mobile phase
- Thermally-stable

### Important parameters ?

- Stationary phase composition
- Mobile phase composition
- Temperature and pressure

## Liquid

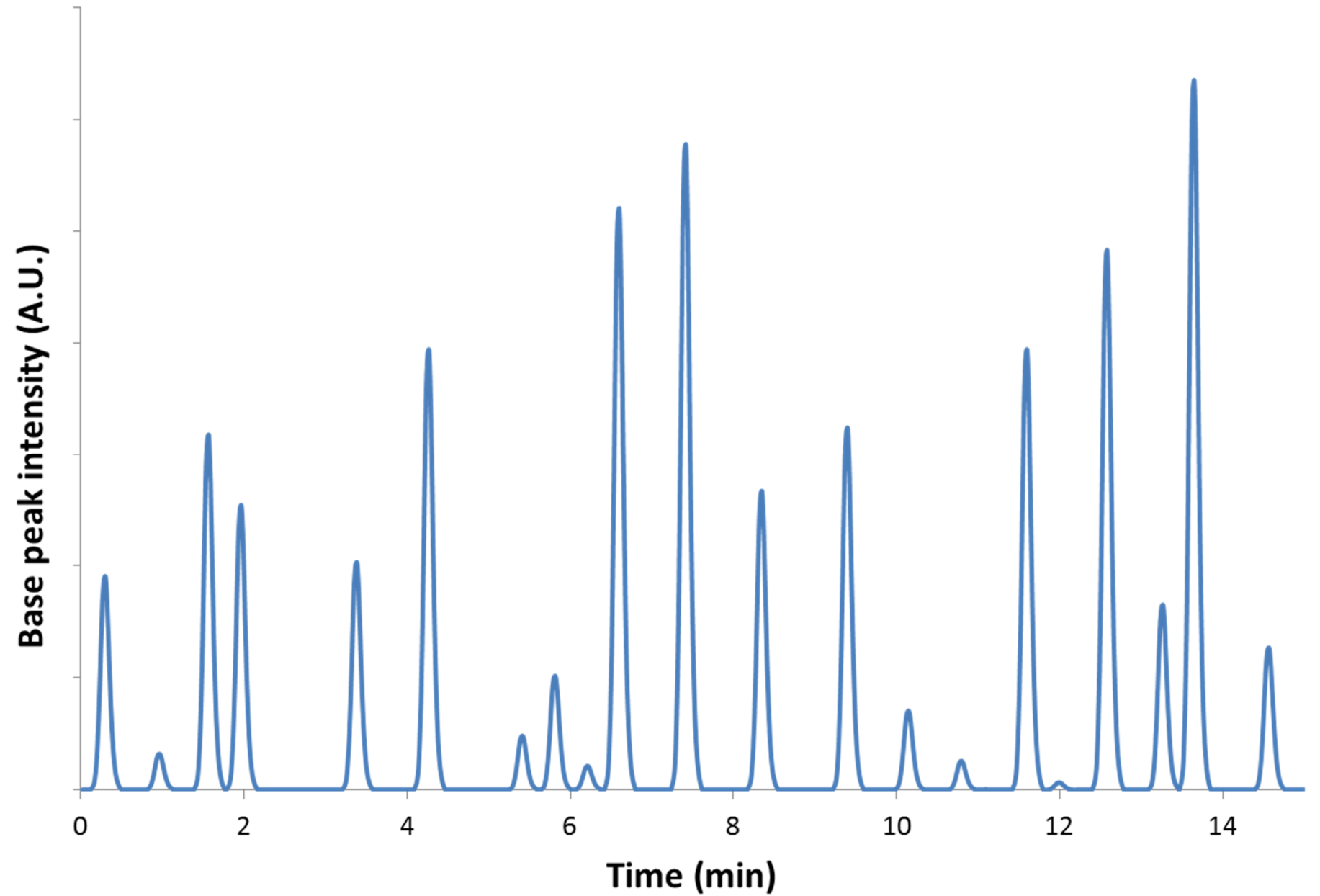
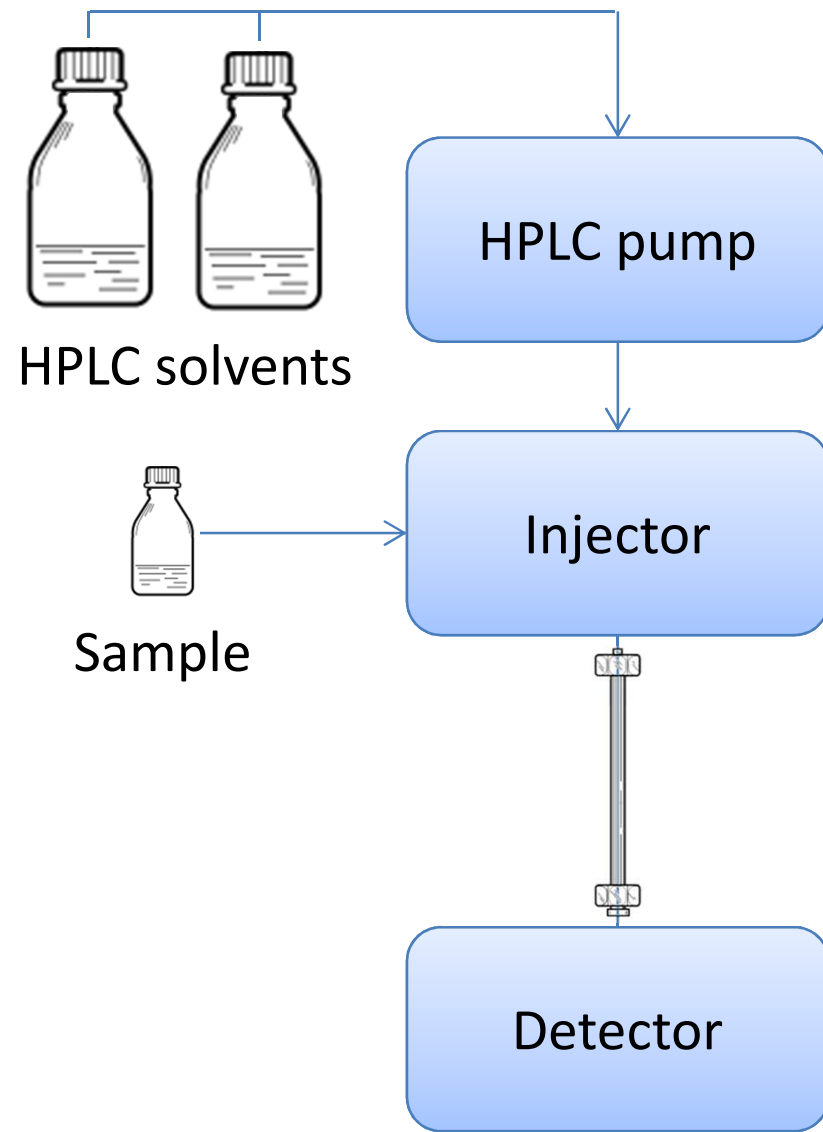
### Analyte ?

- Soluble in mobile phase

### Important parameters ?

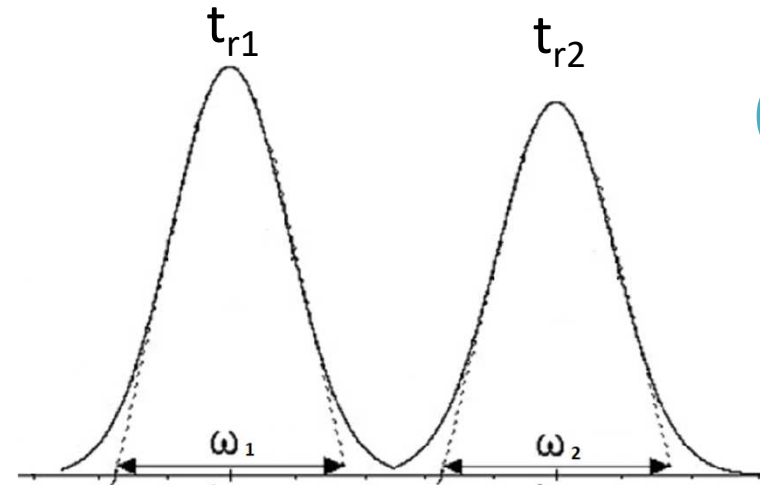
- Stationary phase composition
- Mobile phase composition
- Temperature

# HPLC



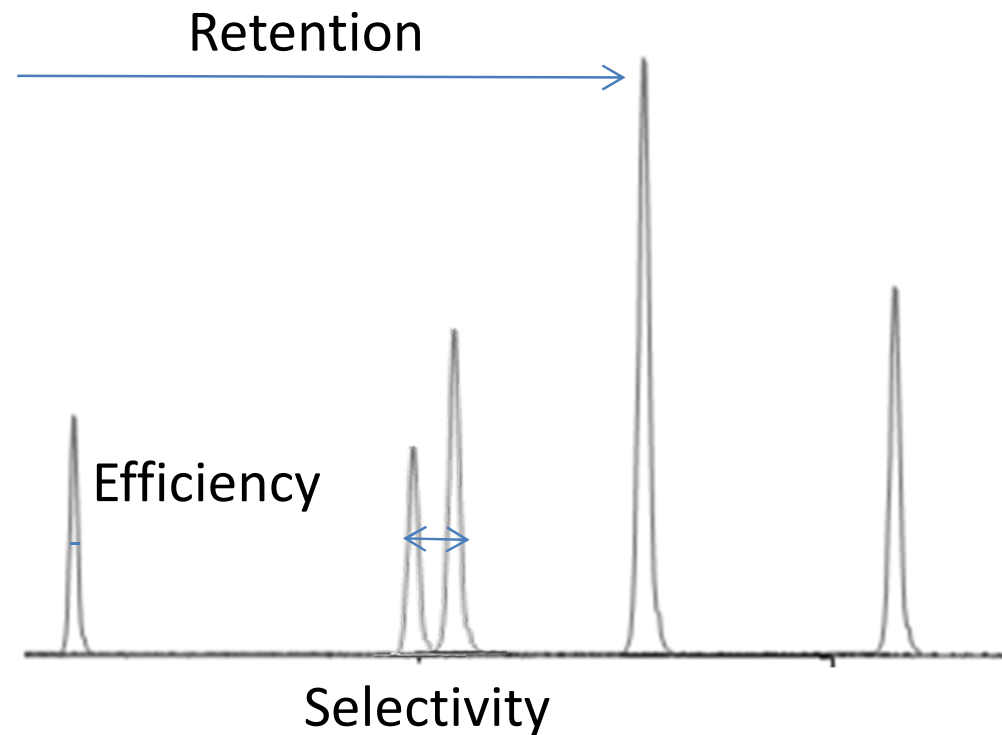
# Resolution

$$R_S = \frac{2(t_{r1} - t_{r2})}{\omega_1 + \omega_2}$$



Optimal:  $1,4 < R_S < 1,6$

$\omega_1 \approx \omega_2$  (Purnell)



$$R_S = \frac{1}{4} \sqrt{N} \times \frac{\alpha - 1}{\alpha} \times \frac{k}{1 + k}$$

Efficiency

Selectivity

Retention

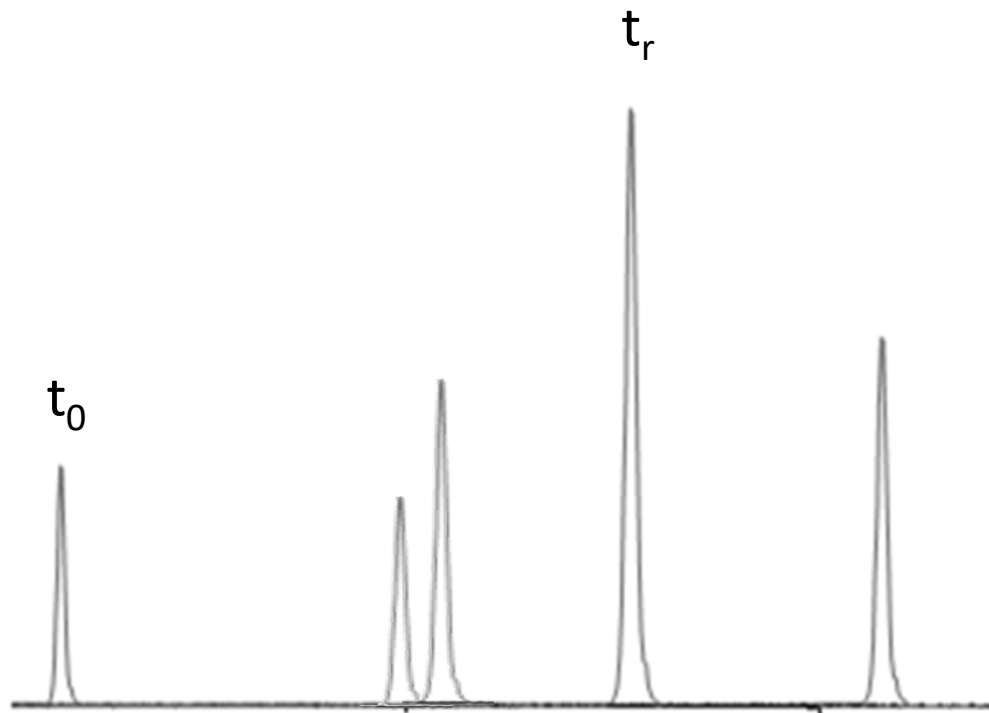


# Retention

$$R_s = 1/4 \sqrt{N} \times \frac{\alpha - 1}{\alpha} \times \frac{k}{1 + k}$$

Characterize the compound retention

$$k = \frac{t_r - t_0}{t_0}$$



Parameters affecting retention:

- mobile phase
- stationary phase

# Selectivity

$$R_S = 1/4 \sqrt{N} \times \frac{\alpha - 1}{\alpha} \times \frac{k}{1 + k}$$

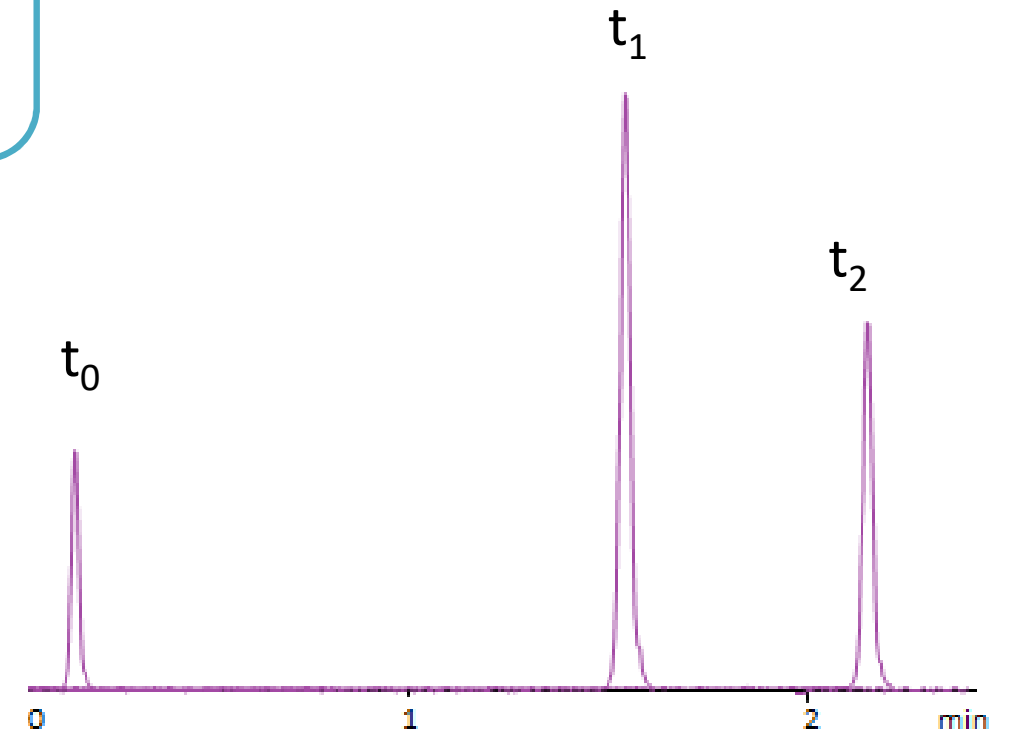
System ability to separate two components

$$\alpha = \frac{t_2 - t_0}{t_1 - t_0} = \frac{k_2}{k_1}$$

$\alpha = 1$  no separation.

Parameters affecting selectivity:

- Mobile and stationary phase
- Temperature

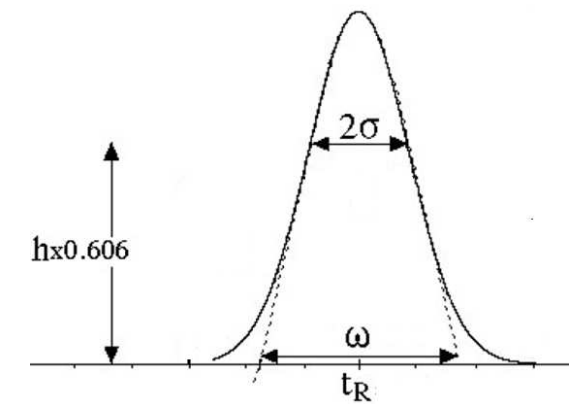


# Efficiency

$$R_s = 1/4 \sqrt{N} \times \frac{\alpha - 1}{\alpha} \times \frac{k}{1 + k}$$

N (plate number) reflects the column performance → measure of the peak dispersion on the column

$$N = 16 \left( \frac{t_r}{\omega} \right)^2 = \left( \frac{t_r}{\sigma} \right)^2$$

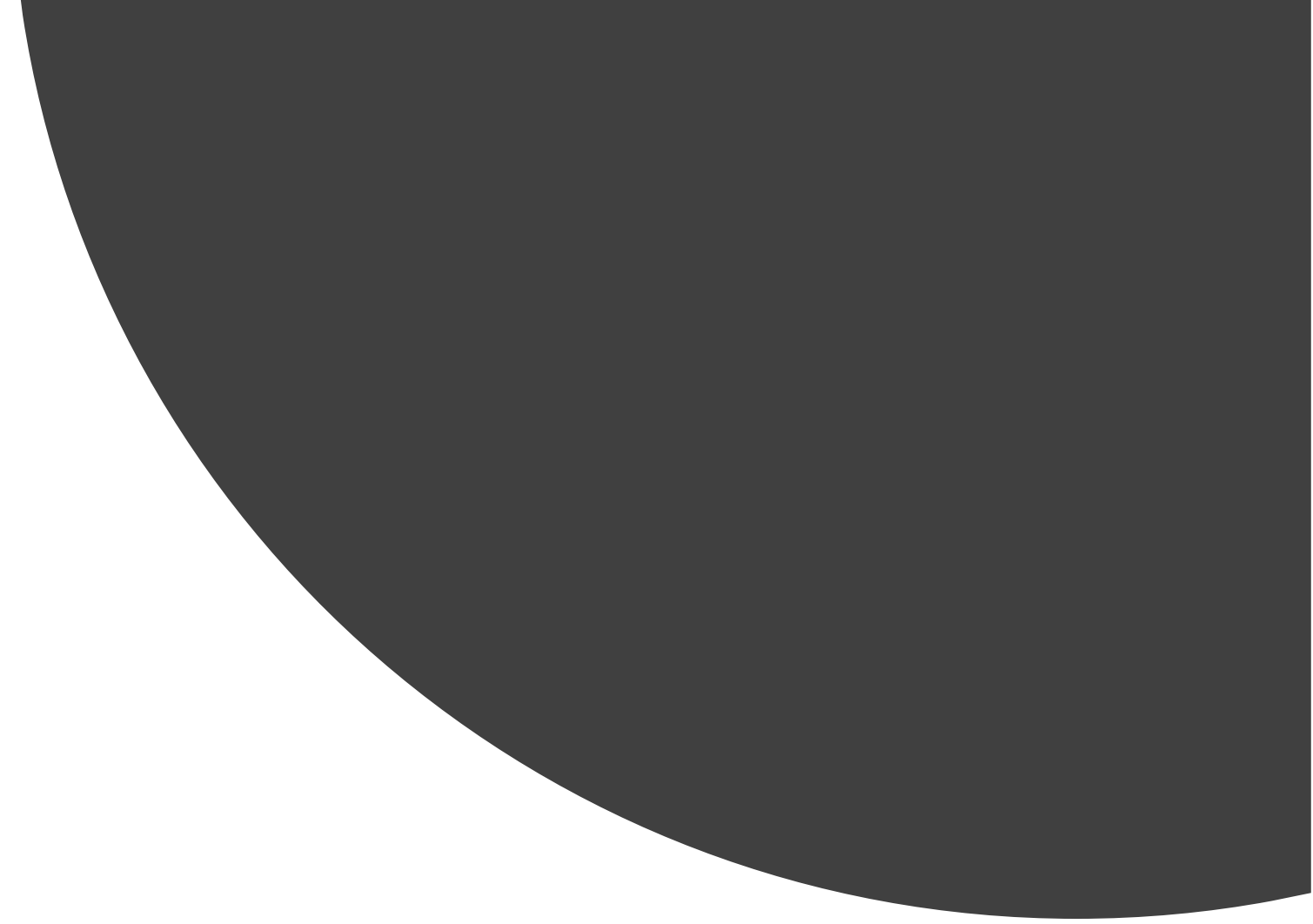
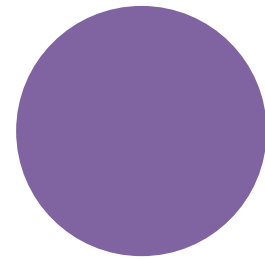
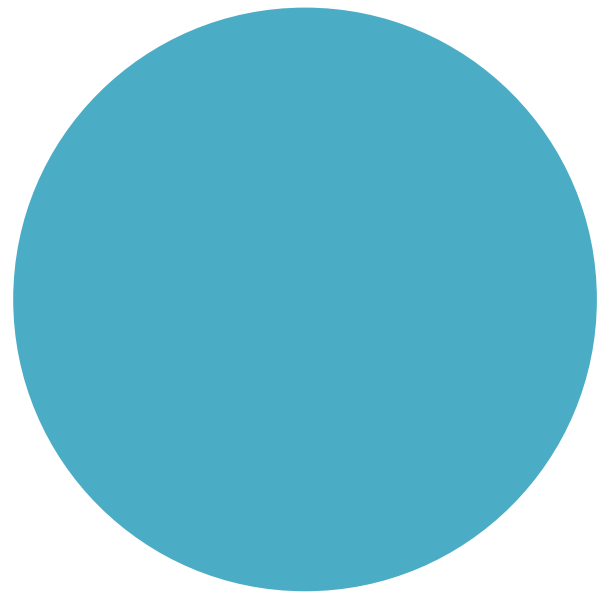


$\sigma$ : standard deviation  
Normal distribution:

$$\omega = 4\sigma$$

Parameters affecting efficiency:

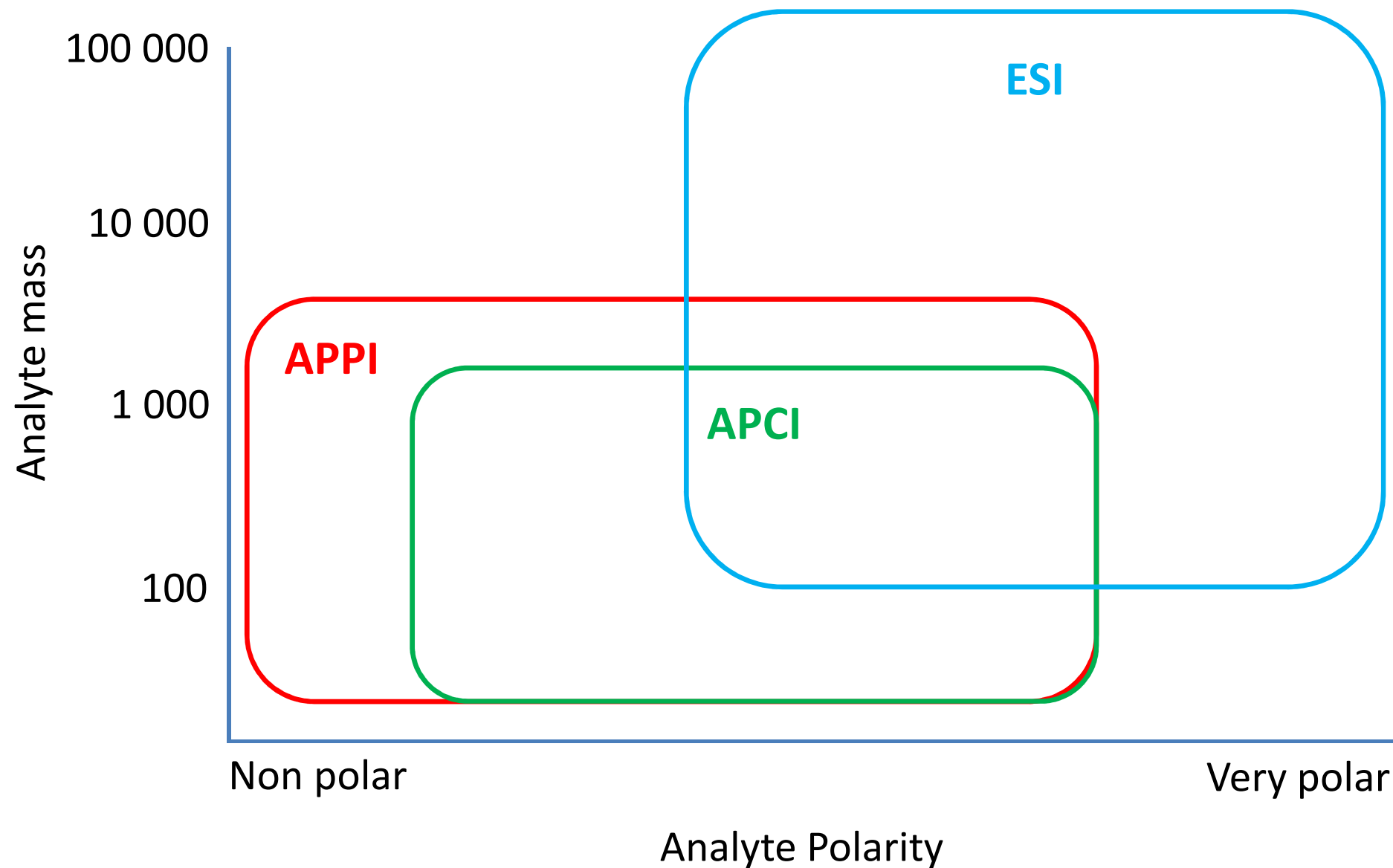
- Column length
- Mobile and stationary phase (granulometry, interactions etc...)
- Analysis time



# LC-FTMS

Sources, acquisition  
frequency, high  
resolution

# Sources



- ESI
  - Polar
  - High mass
- APCI
  - Low polarity
  - Small mass
- APPI
  - Non polar
  - Small mass

# Electrospray ionization (ESI)

0,8 to 6 kV

Nebulizing gas

Drying gas

Mass spectrometer

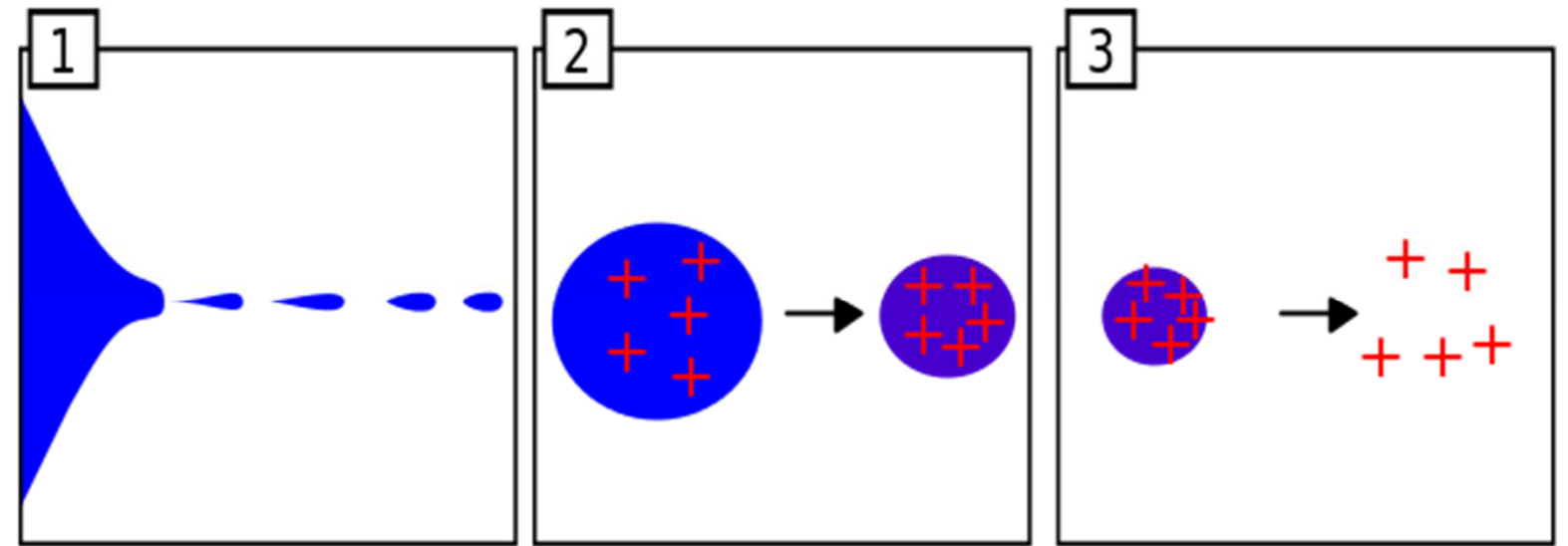
Solvent spray

Ions

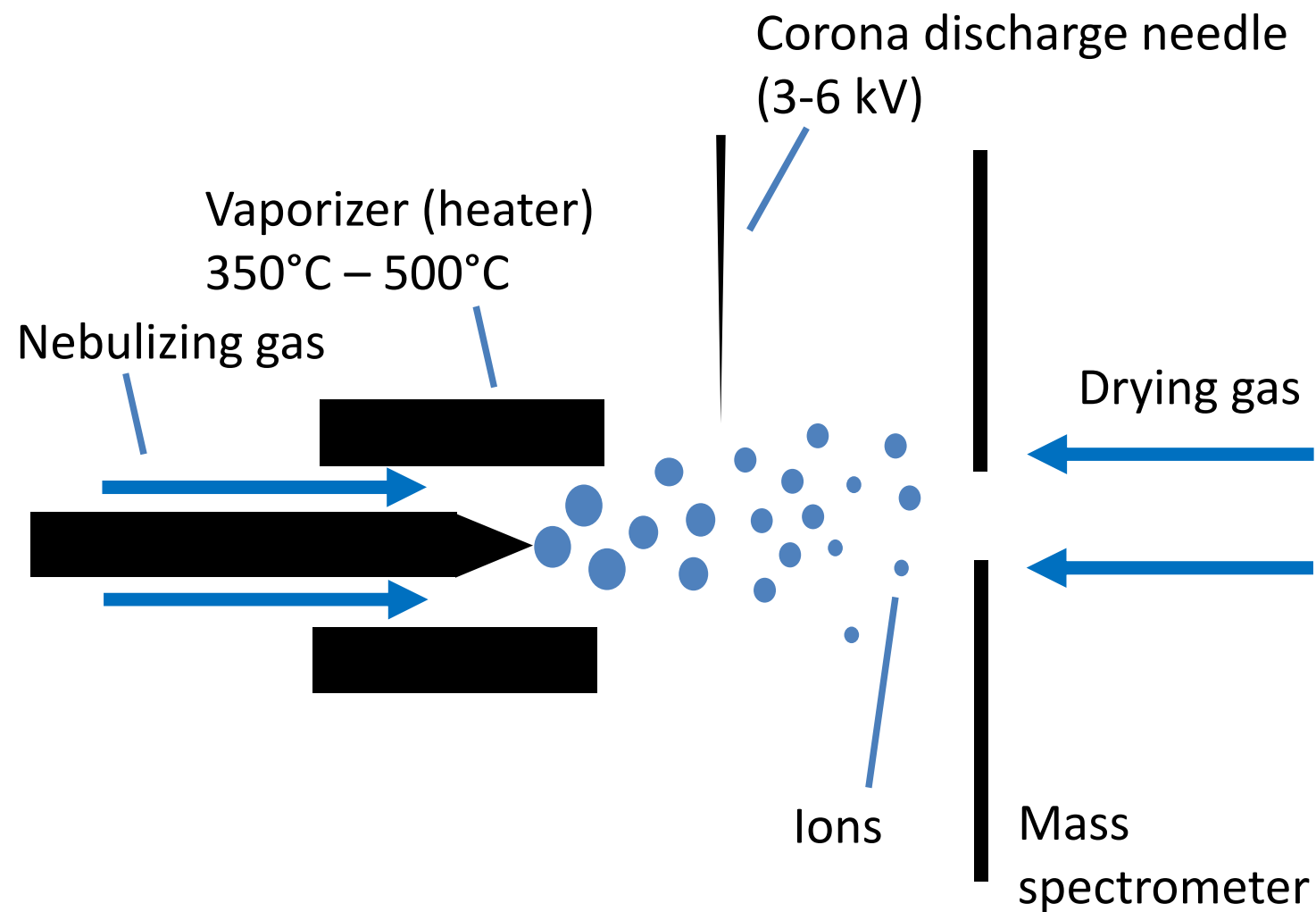
1980 : John Fenn

Wide range of charge states  
Soft ionization technique

Salt adducts



# Atmospheric pressure chemical ionization (APCI)



Reagent ion formation:



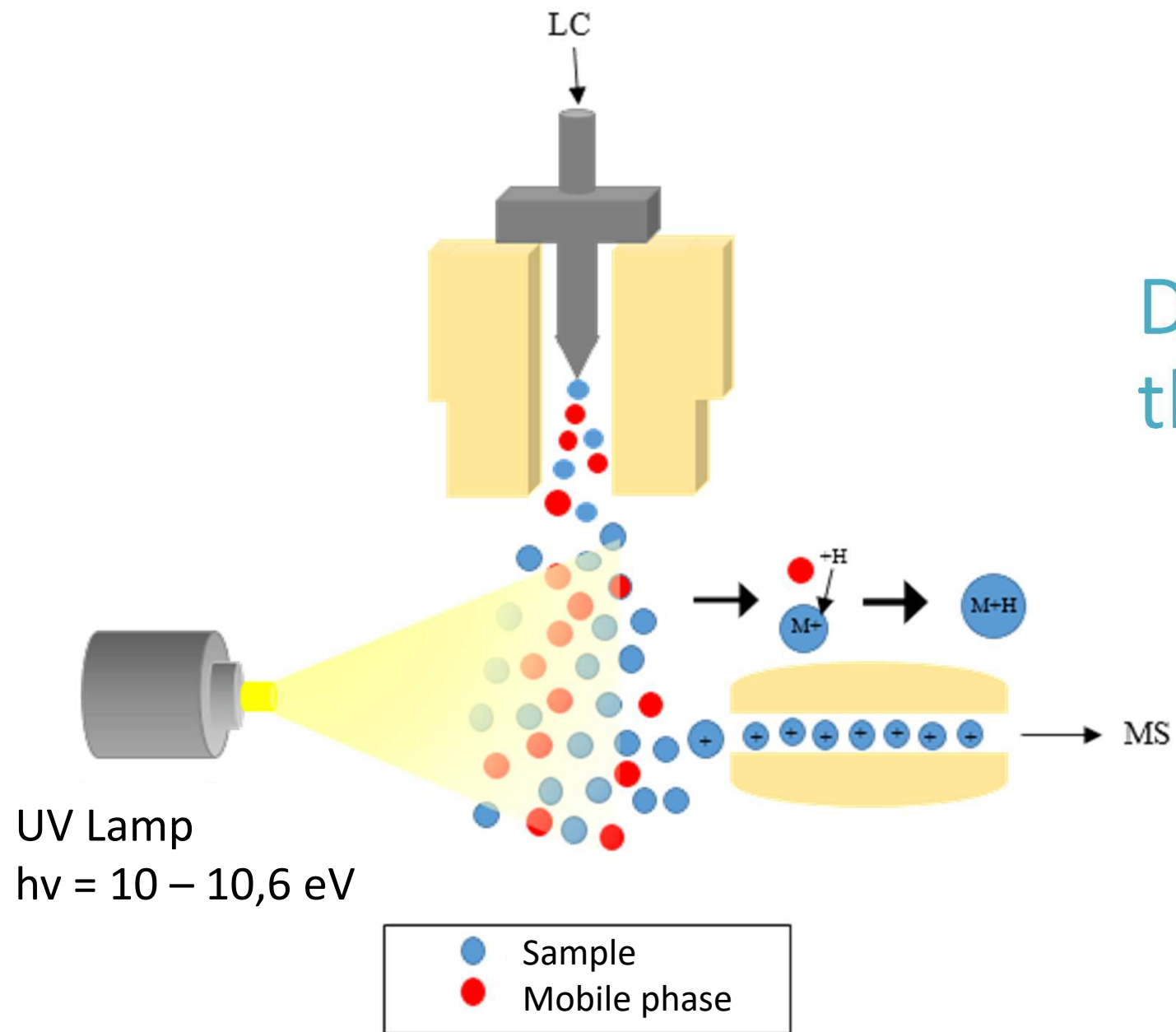
Ionization of product ions:



Declustering:



# Atmospheric pressure photoionization (APPI)



Dopant with lower ionization energy than the sample. (10% to 30% v/v)



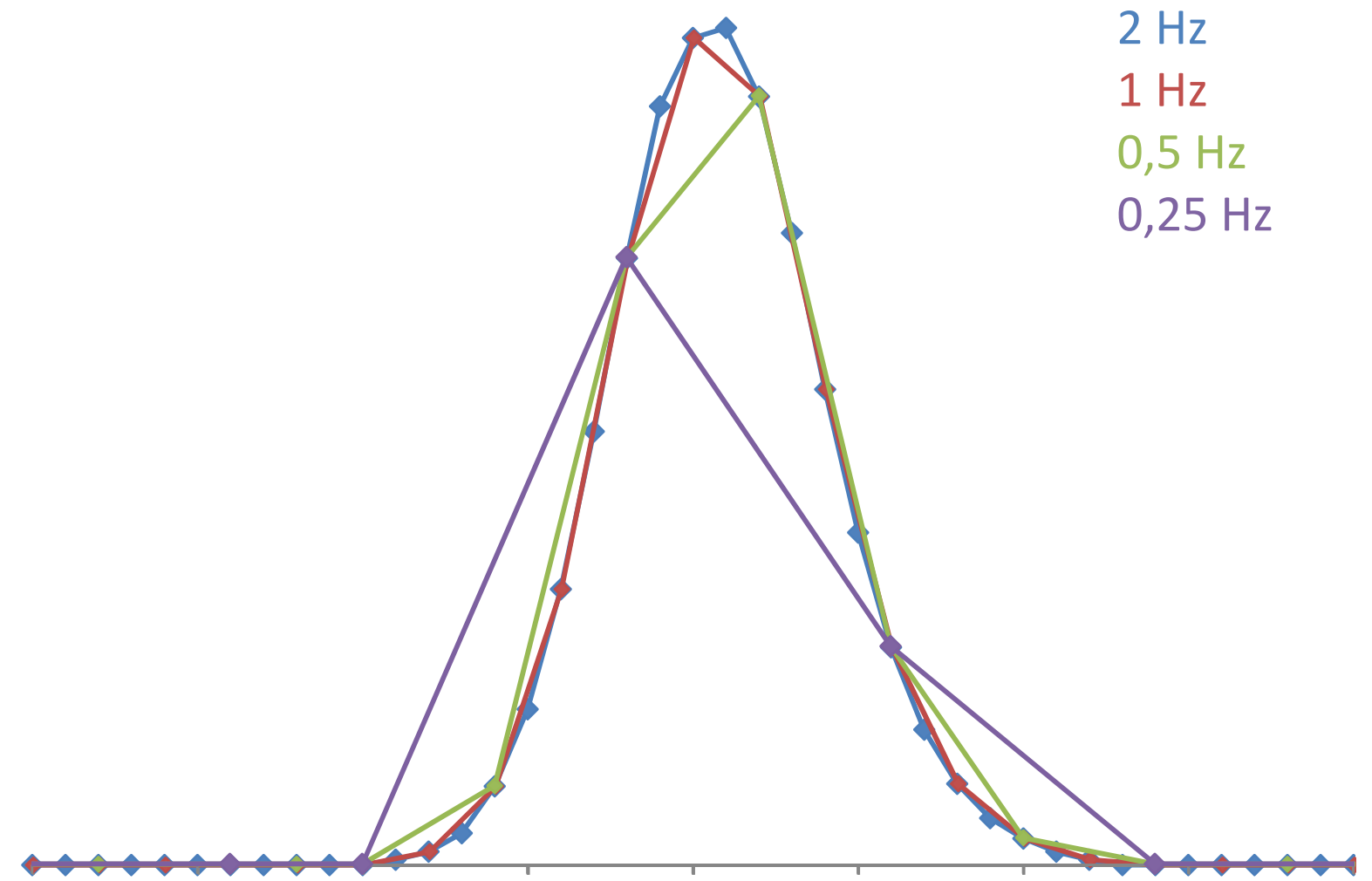


# Comparison ESI / APCI / APPI

	ESI	APCI	APPI
<b>High polarity (drugs, peptides...)</b>	+++	++	++
<b>Medium polarity (steroids, pesticides, lipids...)</b>	+	++	++
<b>Low polarity (PAH...)</b>	--	-	+
High flow rates	+	+++	+
Low flow rates	+++	-	+++
Immunity to ion suppression	--	+	++
Linear dynamic range and quantitation accuracy	+	+++	+++
Reverse phase performance	+++	++	++
Normal phase performance	-	+	+++
<b>Thermally unstable compounds</b>	++	-	+

# Acquisition frequency

Higher scan frequency for better chromatographic resolution



Mass resolving power depends on scan rate

# Minimum resolution?

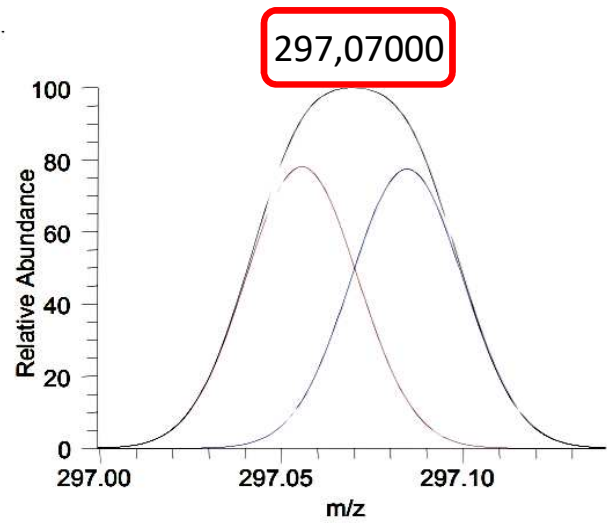
Complex mixture spiked with 151 pesticides, veterinary drugs, mycotoxins and plant toxins (10 to 250 ng/g)

- High selectivity for every singly compound in the mixture
- High specificity and sensitivity
- Confirmation of compound identity
- Accurate quantification of analytes found

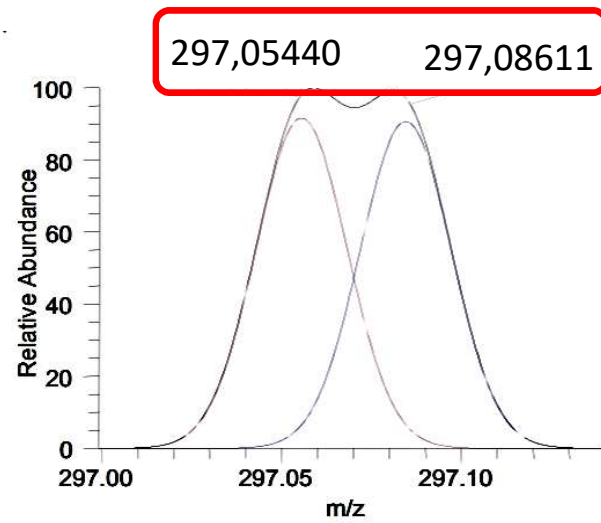
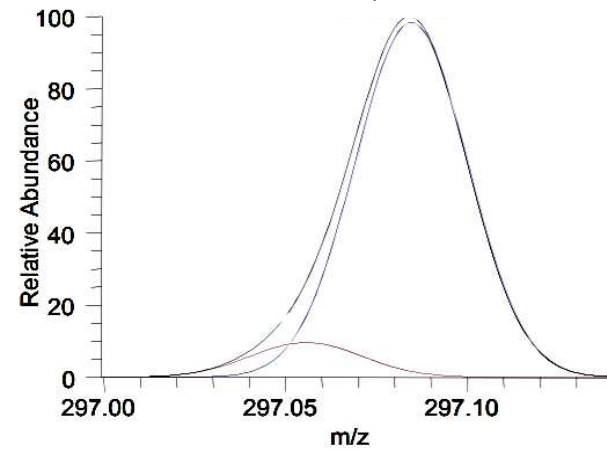
→ Minimum of resolution?



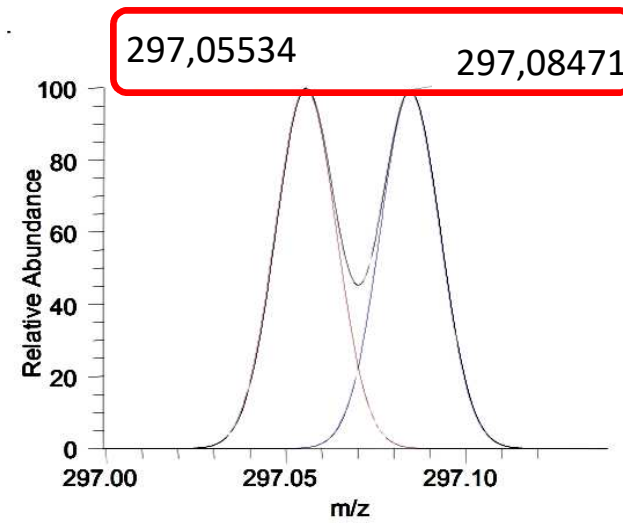
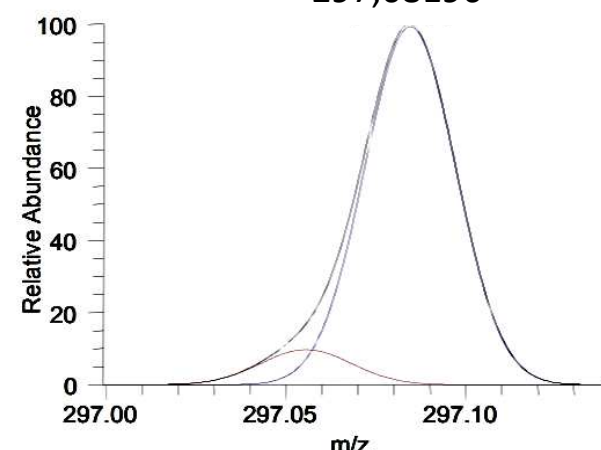
# Minimum resolution?



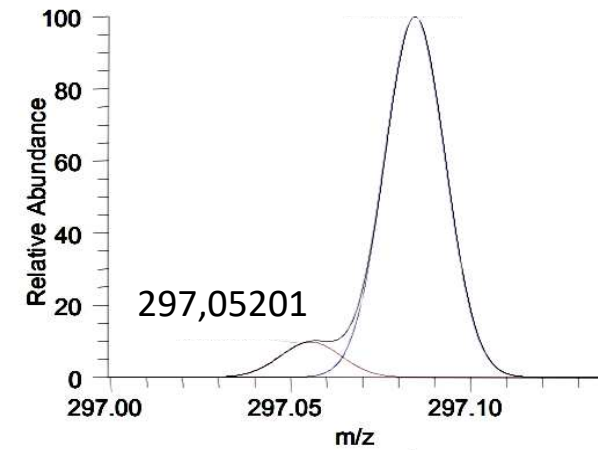
R= 8200  
297,08196



R= 10000  
297,08196



R= 15000  
297,08407



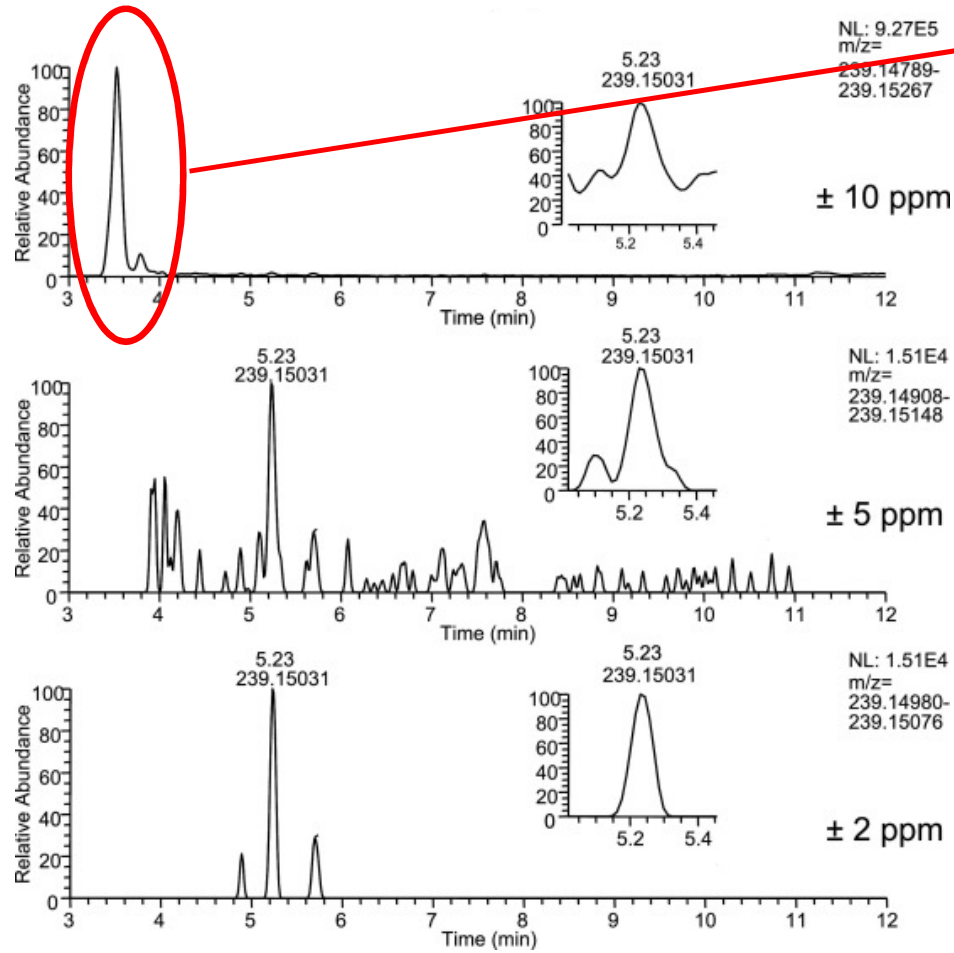
1:1

→ Influence of the resolving power on the mass accuracy

1:10

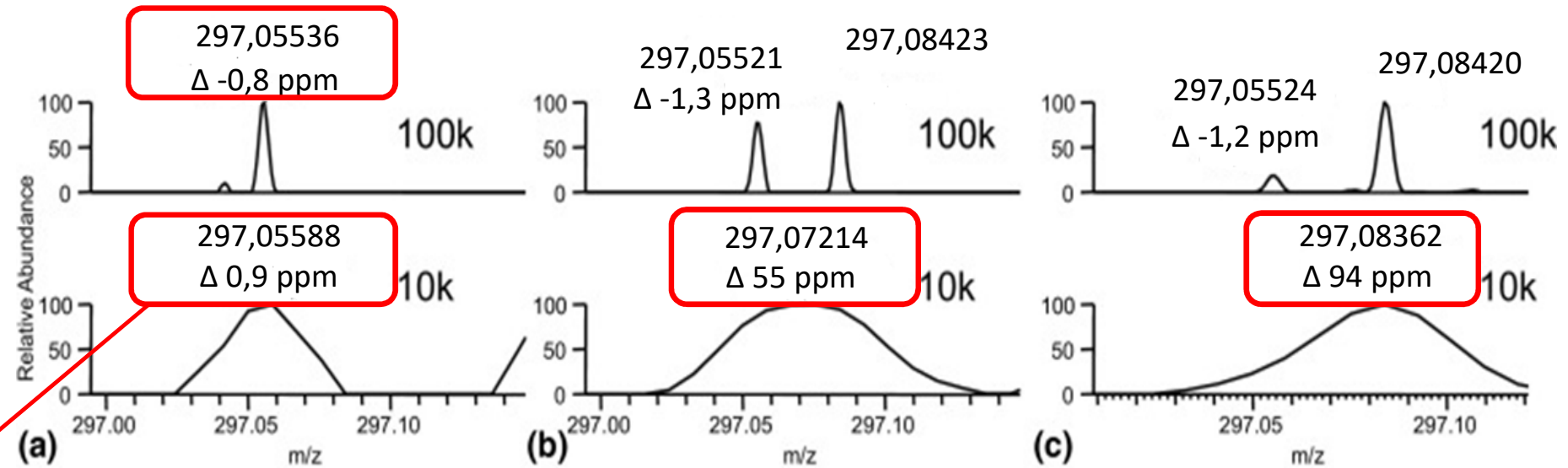
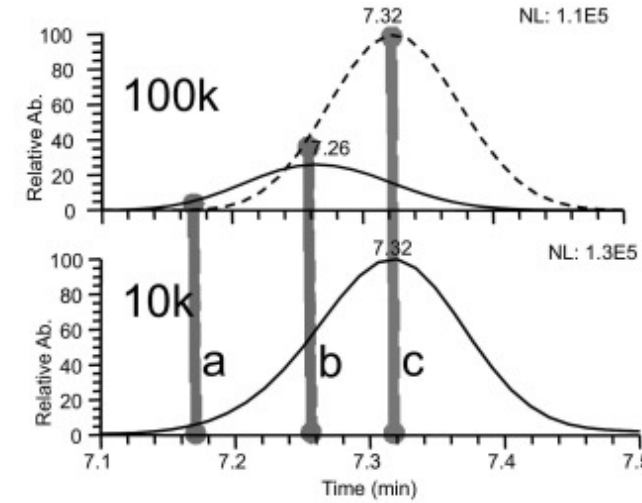
LC (RP C<sub>18</sub>) Orbitrap  
10,000 to 100,000  
(FWHM @ *m/z* 200)  
10 Hz to 1 Hz

# Minimum resolution?



EIC of pirimicarb in horses feed matrix  
(RT = 5,23 min)  
Resolution @ 100,000

Matrix interference



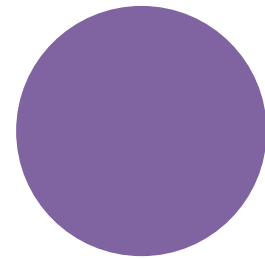
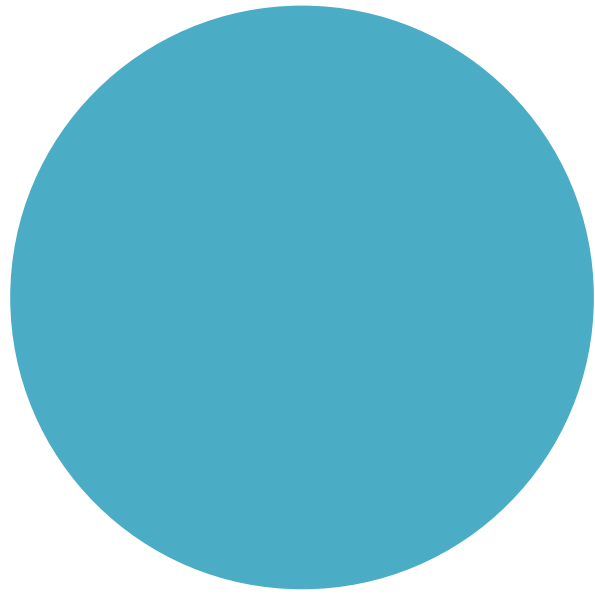
Mass accuracy

Imazalil (MH<sup>+</sup> 297.05560, C<sub>14</sub>H<sub>14</sub>Cl<sub>2</sub>N<sub>2</sub>O, RT = 7,26 min)  
 Flunixin (MH<sup>+</sup> = 297.08454, C<sub>14</sub>H<sub>11</sub>F<sub>3</sub>N<sub>2</sub>O<sub>2</sub>, RT = 7,32 min)

# Minimum resolution?

Resolution	% of 151 analytes				
	Mass deviation (ppm)				
	< 2	2-5	5-10	10-25	>25 / ND
<b>Honey</b>					
10,000	83	12	5	0	0
25,000	100	0	0	0	0
50,000	100	0	0	0	0
100,000	100	0	0	0	0
<b>Animal feed</b>					
10,000	22	25	16	20	17
25,000	41	17	26	2	14
50,000	93	4	1	0	2
100,000	99	0	0	0	1

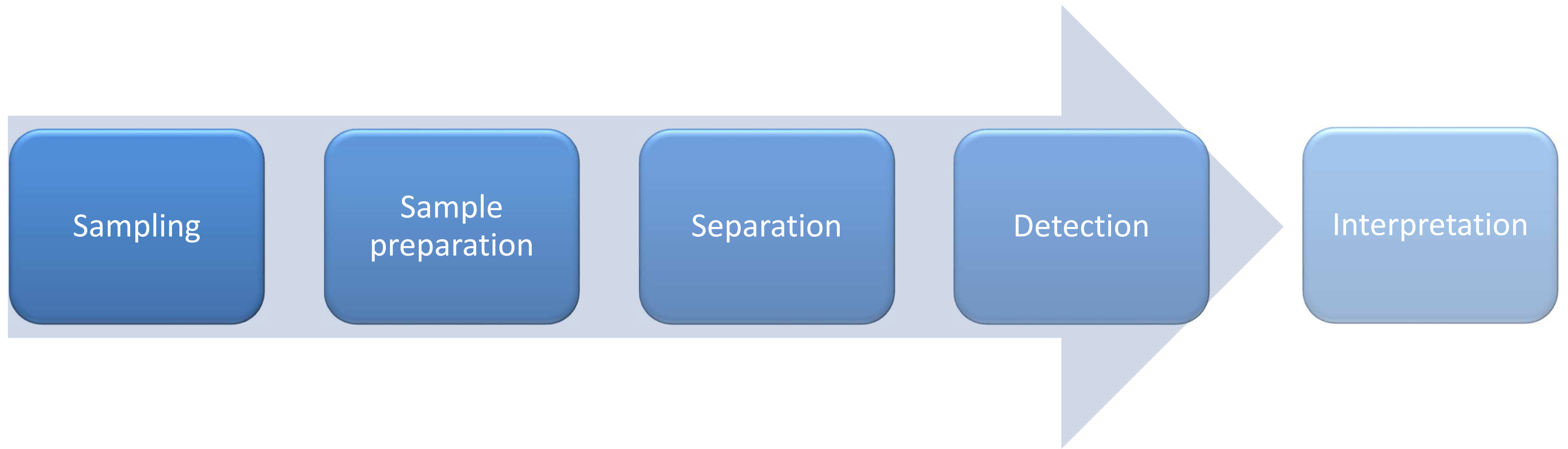
“In **highly complex extracts**, a resolving power of **35,000–50,000 or even 70,000–100,000 is required.**”



# POLLUTANTS ANALYSIS

Sampling, preparation,  
analysis

# Analytical process





# Sampling

- Soils
  - Excavation
  - Drilling
  - Core sampling...
- Water
  - Water column
  - Rosette sampler
  - Passive sampler...
- Air
  - Bubbling
  - Filters
  - Adsorbants...



# Matrix effect

Matrix influence on the analyte

Retention time shift, ion suppression, mass accuracy...



Reduce component of matrix (dilution, sample preparation...)

Improve chromatography (gradient, mobile and stationary phases...)

Change ionisation mode (positif/negatif or ESI/APPI/APCI)

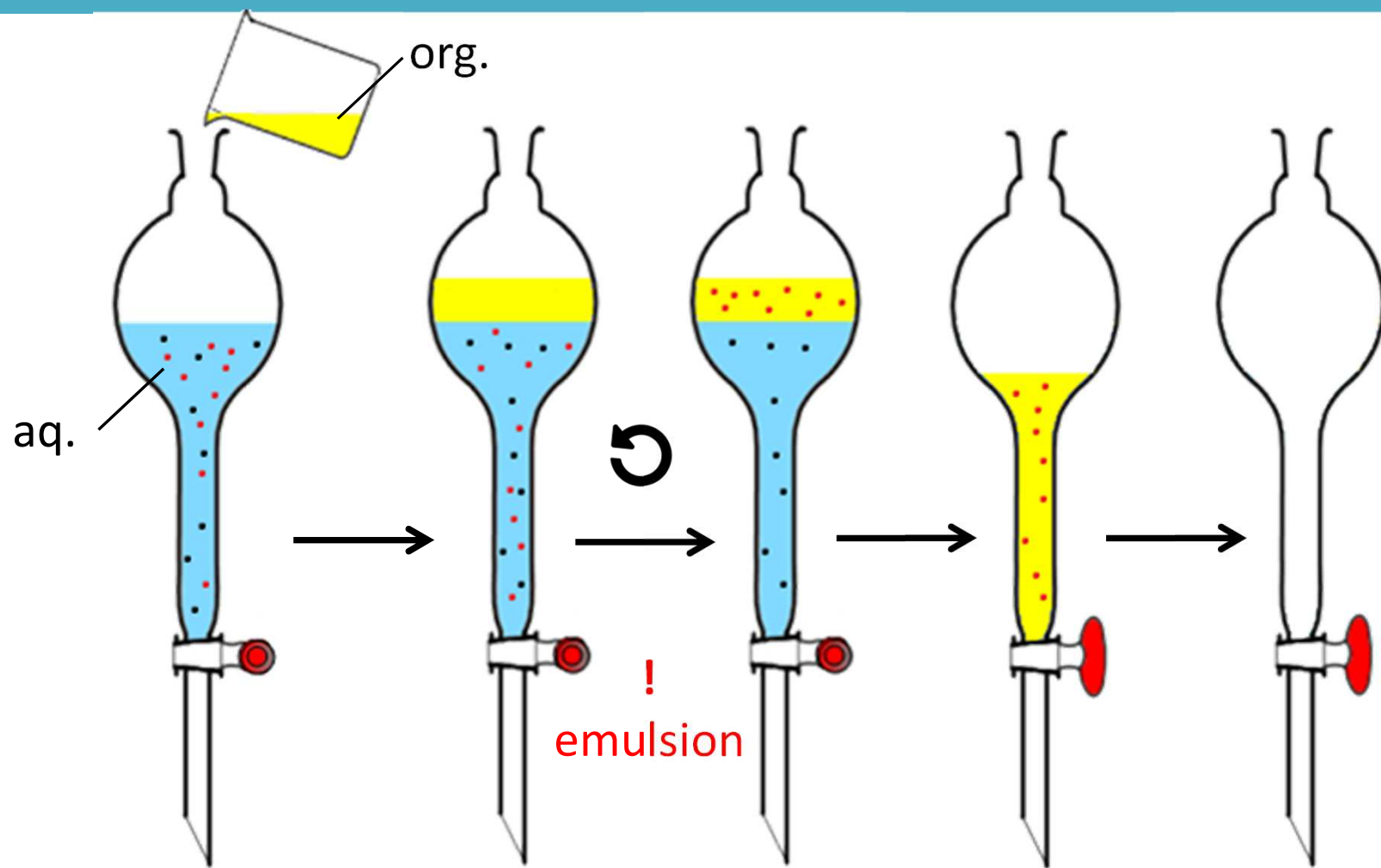
Use of reference matrix and standards

# Sample preparation

- Liquid sample
  - Liquid-liquid extraction (LLE)
  - Solid phase extraction (SPE)
  - Dilution...
- Solid sample
  - Ultrasonic assisted extraction (UAE)
  - Microwave assisted extraction (MAE)
  - Liquid-solid extraction (LSE)...



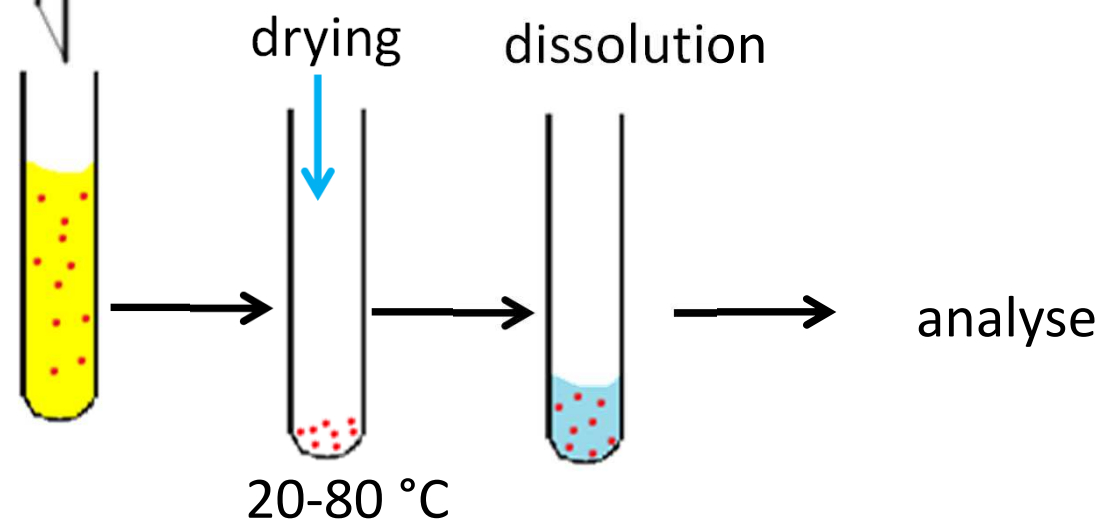
# Liquid-liquid extraction (LLE)



Separation of compounds based on relative solubility in immiscible liquids.

Variant:

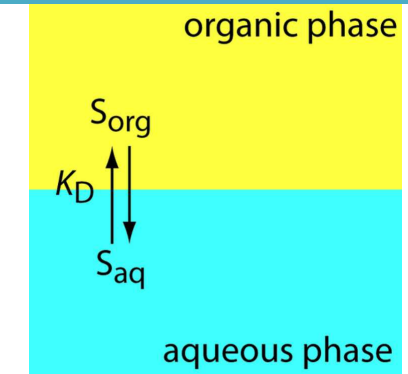
- Dispersive liquid liquid microextraction (DLLME)
- Single drop microextraction (SDME)



# Extraction efficiency

$K_D$  : partition coefficient

$$K_D = \frac{[S]_{org}}{[S]_{aq}}$$

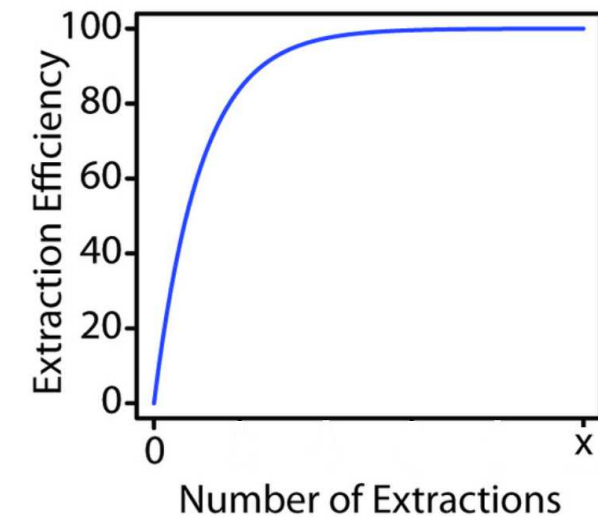


Fraction of solute remaining in aqueous phase after extraction

$$(q_{aq})_1 = \frac{n_{S(aq)1}}{n_{S(aq)initial}} = \frac{[S]_{aq} * V_{aq}}{[S]_{org} * V_{org} + [S]_{aq} * V_{aq}} = \frac{V_{aq}}{K_D * V_{org} + V_{aq}}$$

Extraction efficiency  $X$  for  $n$  extractions:

$$X = 1 - (q_{aq})_n = 1 - \left( \frac{V_{aq}}{K_D * V_{org} + V_{aq}} \right)^n$$



# Example

$$K_D = 5$$

$$V_{aq} = 50 \text{ mL}$$

$$V_{org} = 15 \text{ mL}$$

$$n = 1$$

$X$  ?

$$X = 1 - \left( \frac{V_{aq}}{K_D * V_{org} + V_{aq}} \right)^n$$

$$X = 1 - \left( \frac{50}{5 * 15 + 50} \right)^1 = 0,6$$

$$X = 0,999$$

$V_{org}$  ?

$$V_{org} = \frac{V_{aq} - (1 - X) * V_{aq}}{(1 - X) * K_D}$$

$$V_{org} = \frac{50 - (1 - 0,999) * 50}{(1 - 0,999) * 5} = 9990 \text{ mL}$$

# Example

$$K_D = 5$$

$$V_{aq} = 50 \text{ mL}$$

$$V_{org} = 15 \text{ mL}$$

$$X = 0,999$$

$n$  ?

$$X = 1 - \left( \frac{V_{aq}}{K_D * V_{org} + V_{aq}} \right)^n$$

$$\log(1 - X) = n \log \left( \frac{V_{aq}}{K_D * V_{org} + V_{aq}} \right)$$

$$n = \frac{\log(1 - X)}{\log \left( \frac{V_{aq}}{K_D * V_{org} + V_{aq}} \right)} = \frac{\log(1 - 0,999)}{\log \left( \frac{50}{5 * 15 + 50} \right)} = 7,53$$



$n$  high,  $V_{org}$  small

$$n = 8$$

# Strengths and weaknesses



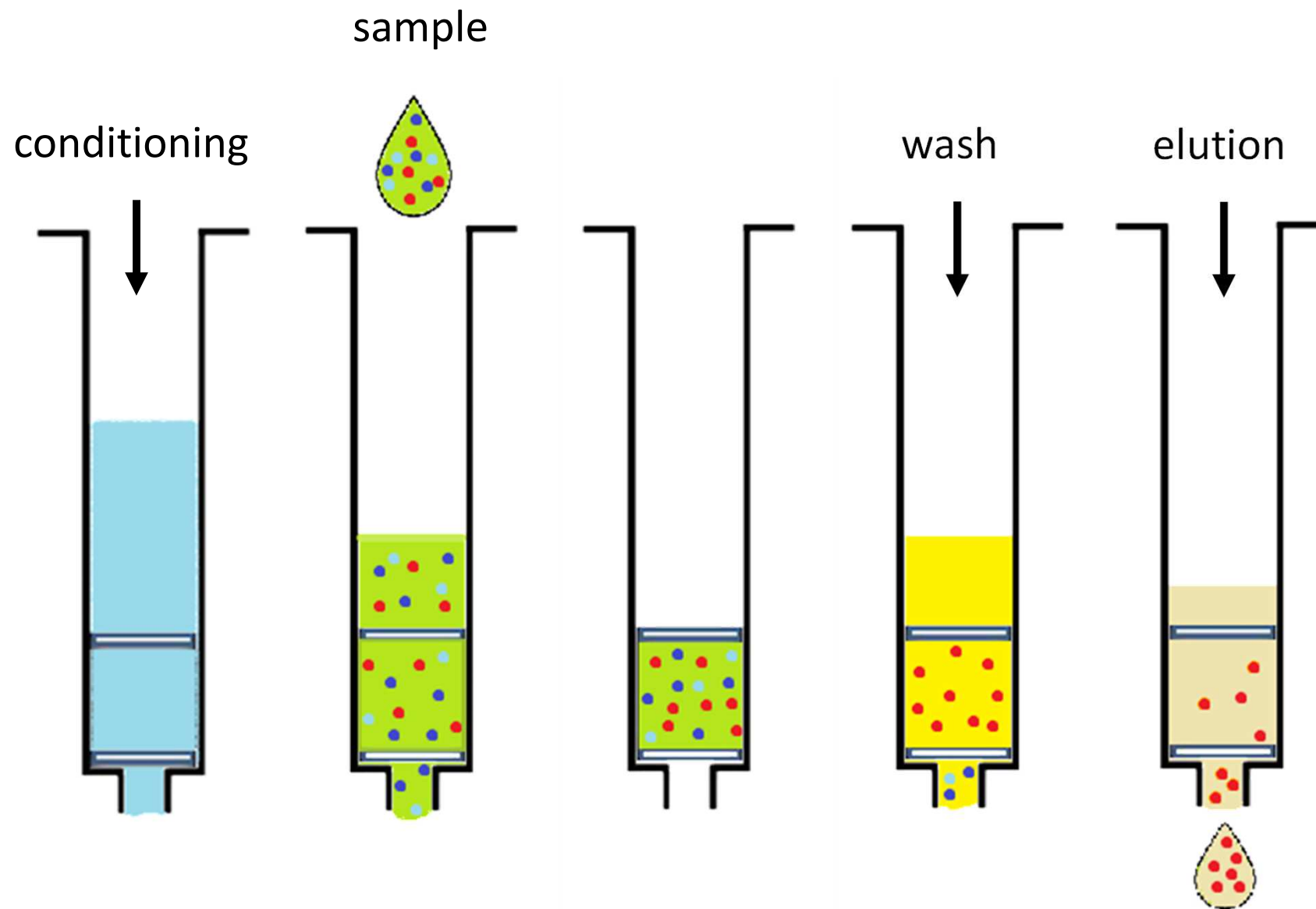
- Wide range of solvents (selectivities and solubilities )
- Easy



- Emulsions
- Time consuming
- Toxicity and volume of solvents
- Number of step



# Solid phase extraction (SPE)



● Analyte

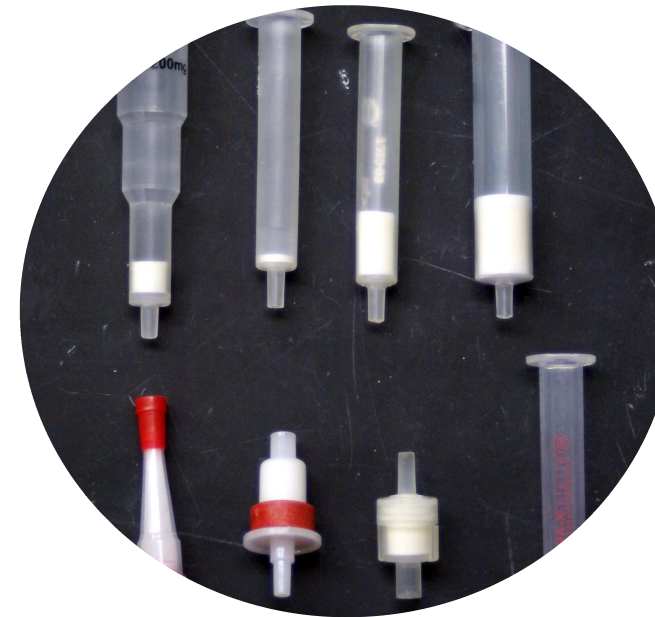
● Contaminants

Comparable with LC:

- Stationary phase
  - Reverse phase, normal phase, Ion Exchange...
- Mobile phase
- Variant:
  - Solid-phase microextraction (SPME)
  - Stir bar sorptive extraction (SBSE)...

# Optimisation

- Sample volume
  - Cartridge capacity
- Nature of adsorbant
  - Hydrophilic or hydrophobic
  - Polar or non-polar...
- Sample pre-treatment
  - pH adjustment
  - Filtration
  - Centrifugation...



# Strengths and weakness

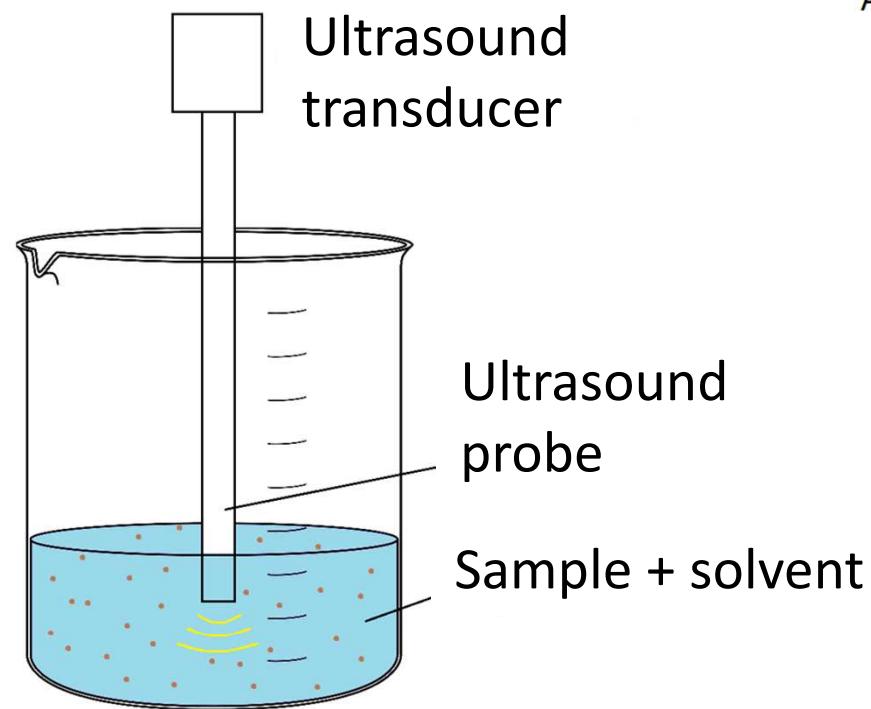


- Time consuming
- Accuracy and precision (flow, volume...)
- Method development

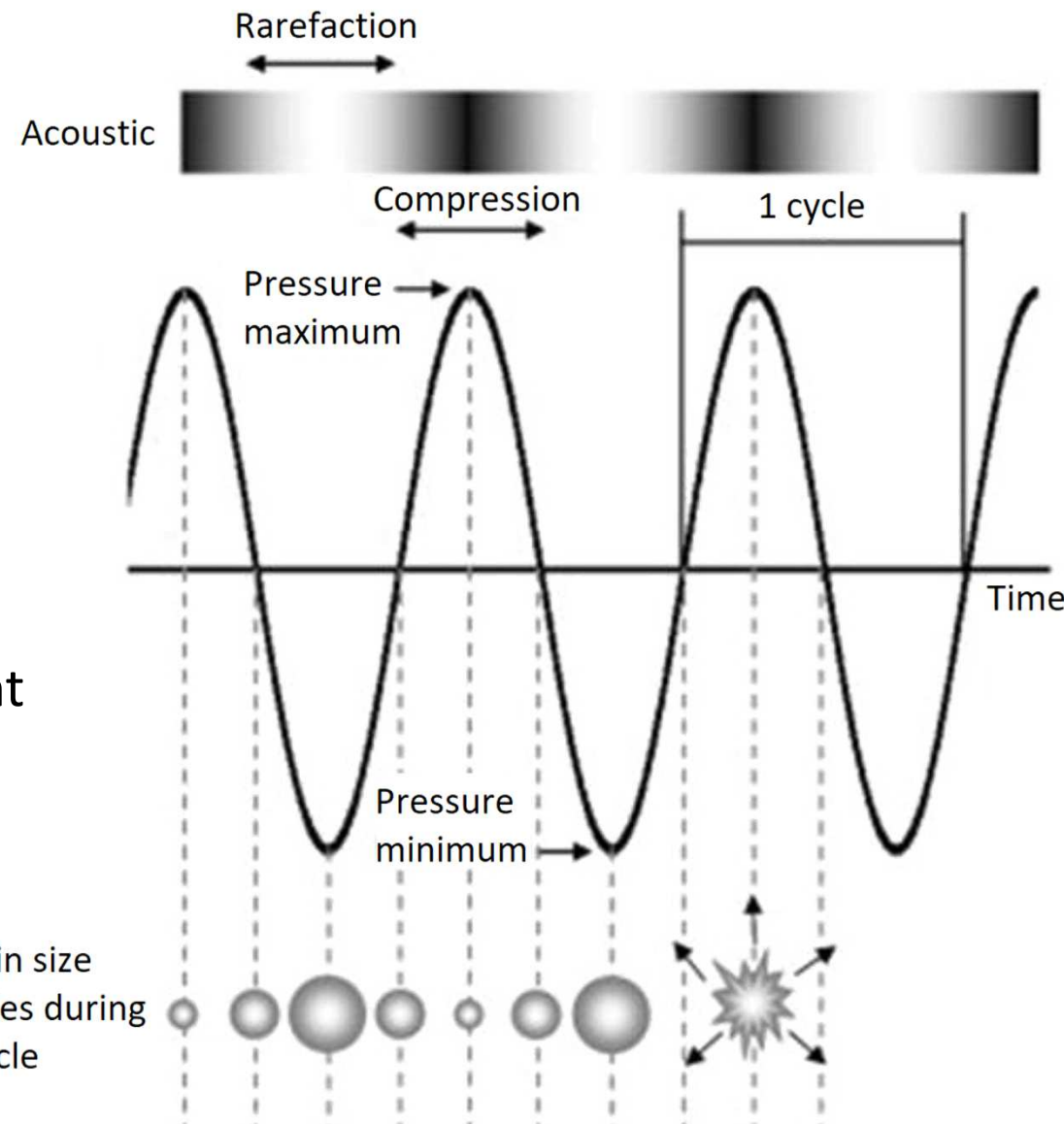


- Automated / On-line
- Large choice of stationary phase, size and extraction solvents
- Preconcentration and extraction

# Ultrasound-assisted extraction (UAE)



Frequency from 20 kHz to 10 MHz



Change in size of bubbles during wave cycle

Low intensity ( $< 1 \text{ W/cm}^2$ ):  
non-destructive analysis

High intensity ( $10 - 1000 \text{ W/cm}^2$ ):

improve efficiency of  
sample preparation

# Strengths and weakness



- Post-treatment (filtration, concentration...)
- Toxicity of extraction solvent



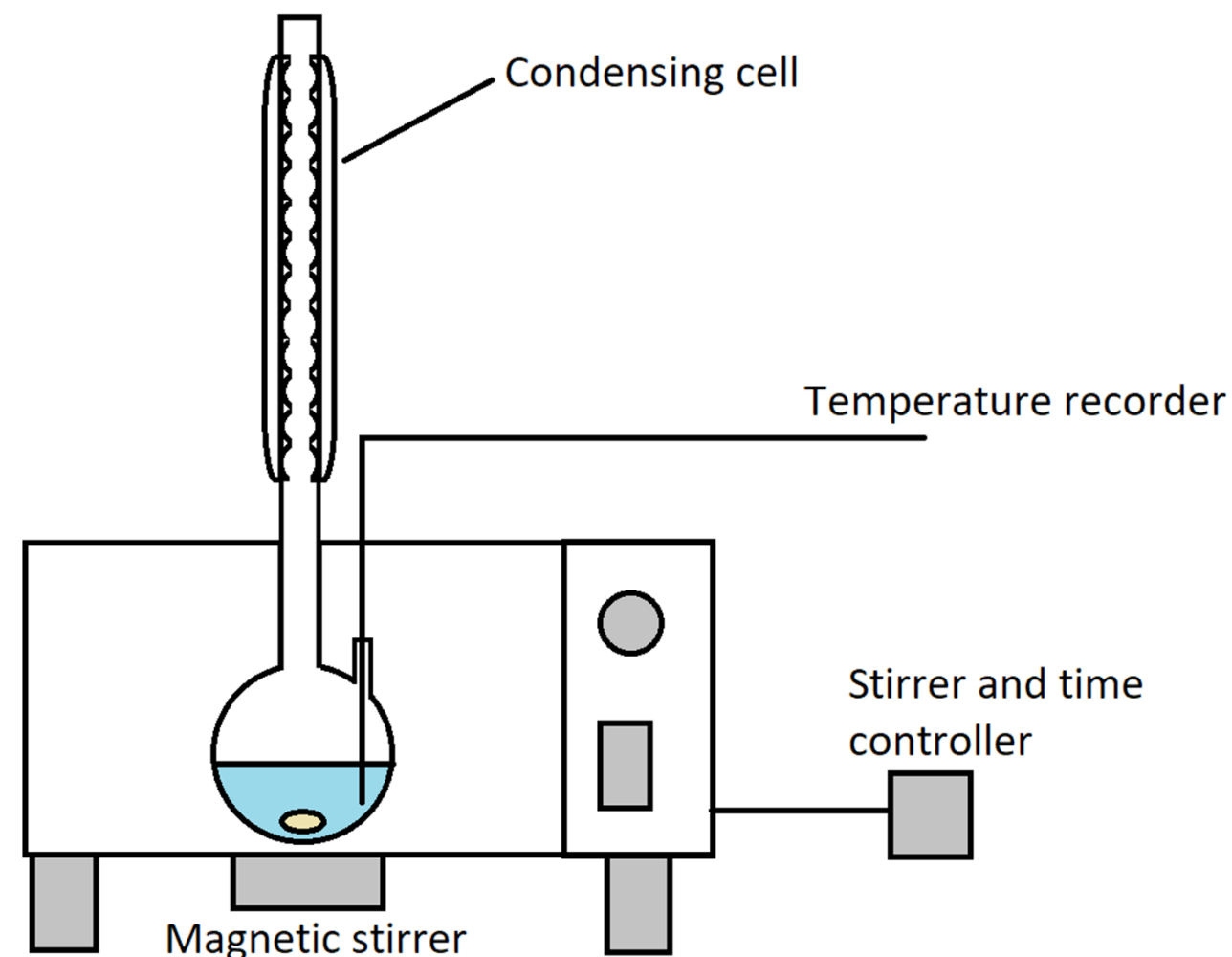
- Fast
- Large scale of application (solids and liquids)
- Unsophisticated instrumentation required

# Microwave assisted extraction (MAE)

Microwave radiation interact with dipole of polar and polarizable material (orient in the changing field direction → heated)

## Variant:

- Focused microwave-assisted extraction (FMAE)
- Pressurized microwave-assisted extraction (PMAE)
- Dynamic microwave-assisted extraction (DMAE)



Microwave region frequencies: 300 MHz to 100 GHz.  
(domestic and scientific ovens operate at 2.45 GHz)

# Strengths and weakness



- Expensive instrumentation
- Re-adsorption possible during cooling step
- Post-treatment



- High extraction rate (rapid heating and elevated temperature)
- Fast and easy
- Automatisation possible (on-line)

# Characterization of marine dissolved organic by LC-FT-ICR (off-line)

## Improvement of DI-FTMS with LC separation

Brine	Drill holes (100 cm of depth)
Edge water	Pumped into clean canister
Under ice water	
Antarctic surface sea water	Rosett sampler connected to CTD (Conductivity, temperature, depth sensor)
Weddell sea deep water	





# Characterization of marine dissolved organic by LC-FT-ICR (off-line)

## Sample preparation:

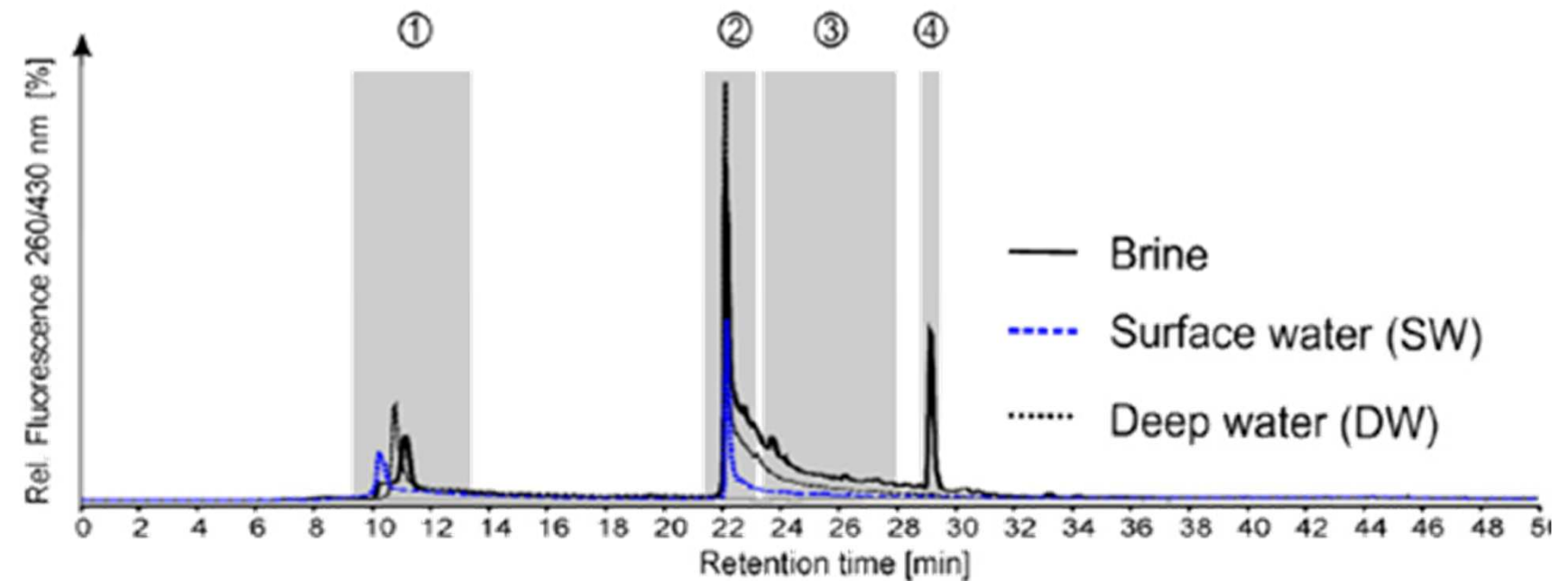
160 L of aqueous sample

Filtration (0,2  $\mu\text{M}$ ) + pH 2 (HCl)

SPE (PPL 5 g, Varian Mega Bond Elut)

## Chromatographic separation:

C18 RP Phenomenex, synergic



# Characterization of marine dissolved organic by LC-FT-ICR (off-line)

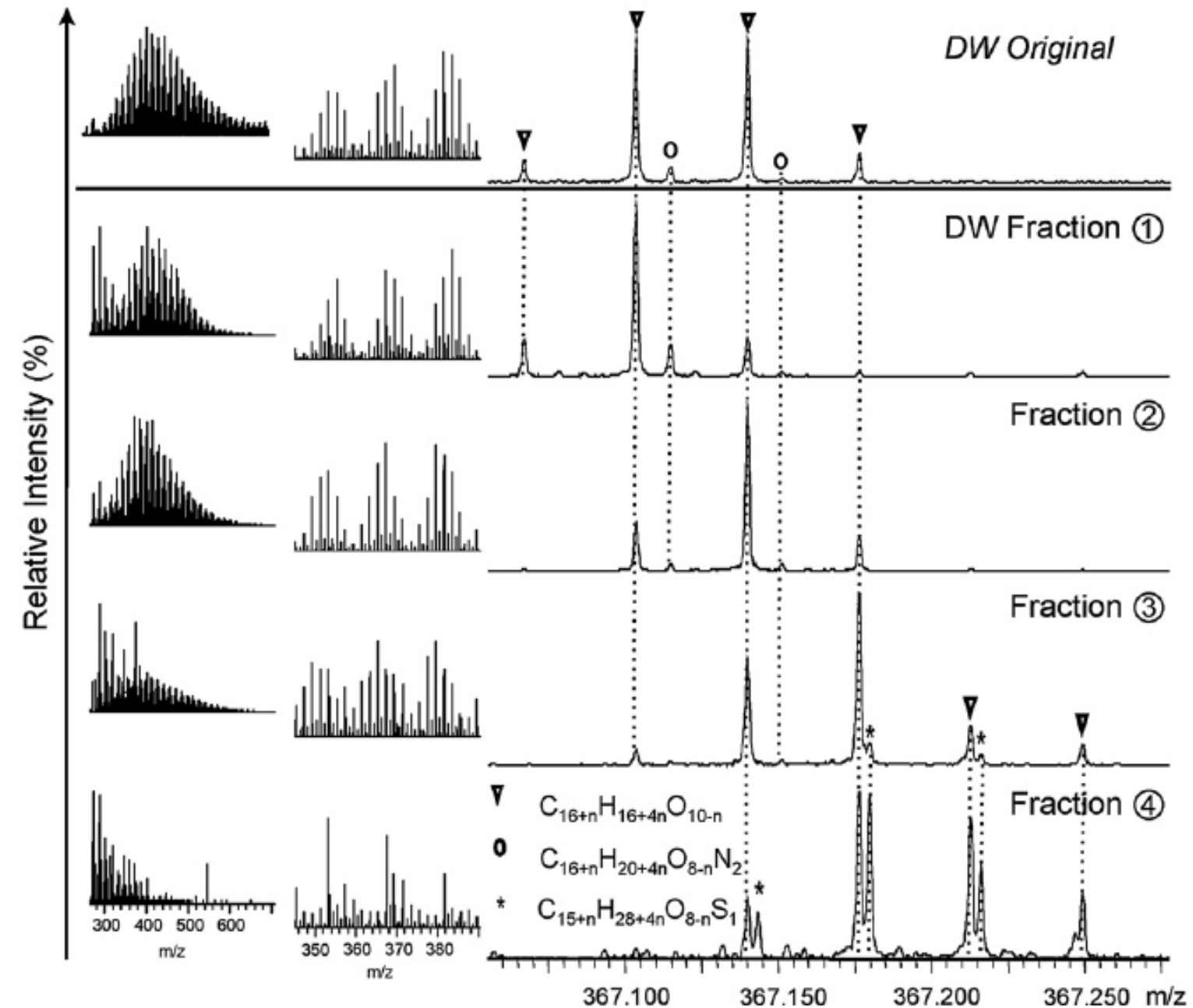
FT-MS analysis:

ApexQe 9,4T (Bruker) ESI neg

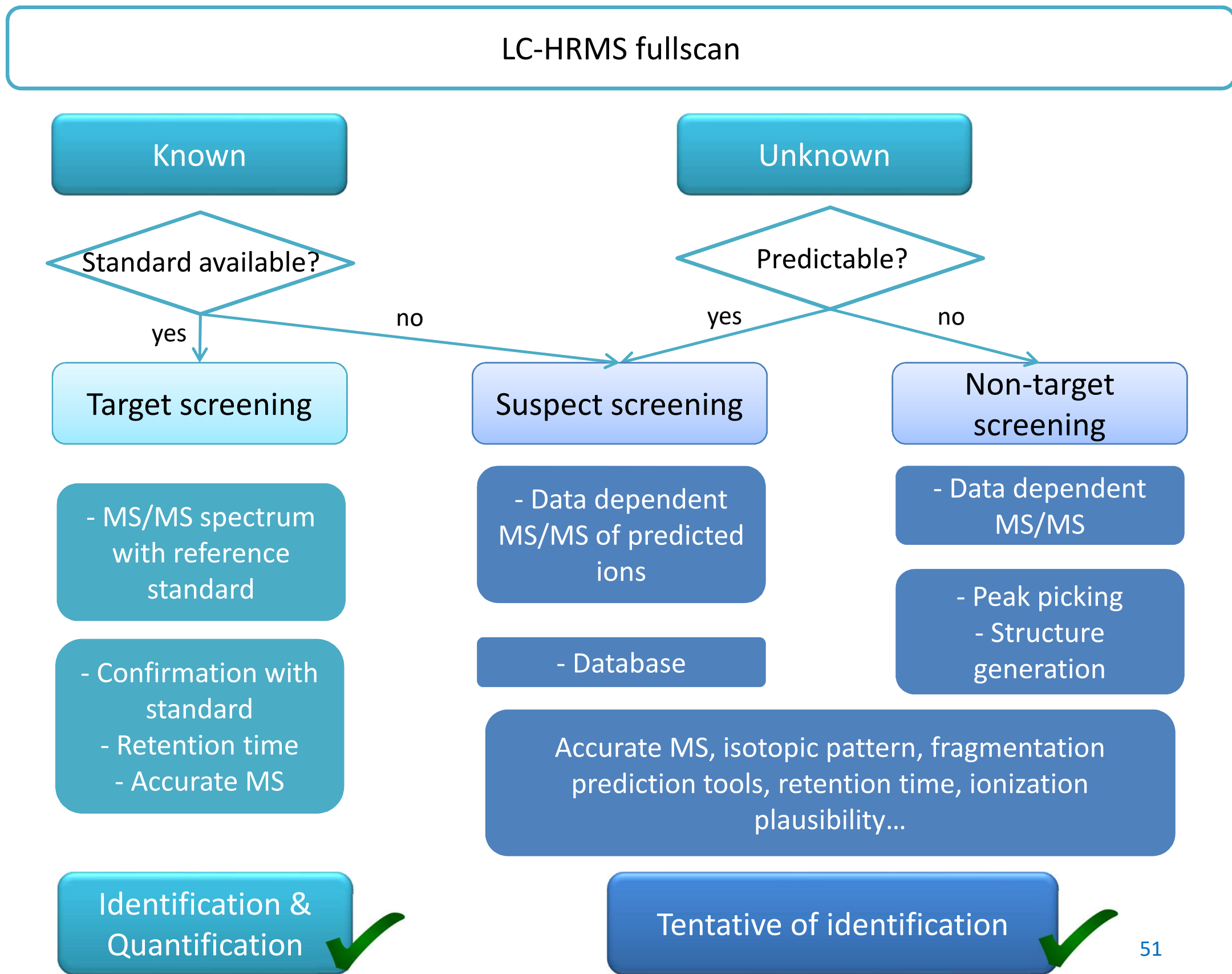
3000 pics per fraction → 75% assigned  
(mass accuracy: 0,5ppm)

→ 70 to 90 % of unique pic

→ Isomers separation



# LC-FTMS analytical strategy



# Target screening

Known products and standards available

## LC-HRMS (fullscan and MS/MS)

- Increase of specificity and selectivity
- all compounds determined simultaneously (fullscan mode)

EU decision 2002/657/EC → 4 identification points (IP)

MS technique	Identification point earned per ion
Low-resolution MS (LR-MS)	1.0
LR-MS <sup>n</sup> precursor ion	1.0
LR-MS <sup>n</sup> transition product	1.5
High-resolution MS (HR-MS)	2.0
HR-MS <sup>n</sup> precursor ion	2.0
HR-MS <sup>n</sup> transition product	2.5

# Target screening of 170 pesticides

Fruits and vegetables extracts spiked with 170 selected pesticides (10  $\mu\text{g}/\text{kg}$  to 500  $\mu\text{g}/\text{kg}$ )

UHPLC-Orbitrap HESI positive:

- R = 17,500; 35,000 and 70,000 (FWHM @  $m/z$  200)  $\rightarrow$  12; 7 and 3 scans per second
- Thermo scientific accucore AqC<sub>18</sub>

Orbitrap had similar quantitative performance than QqQ

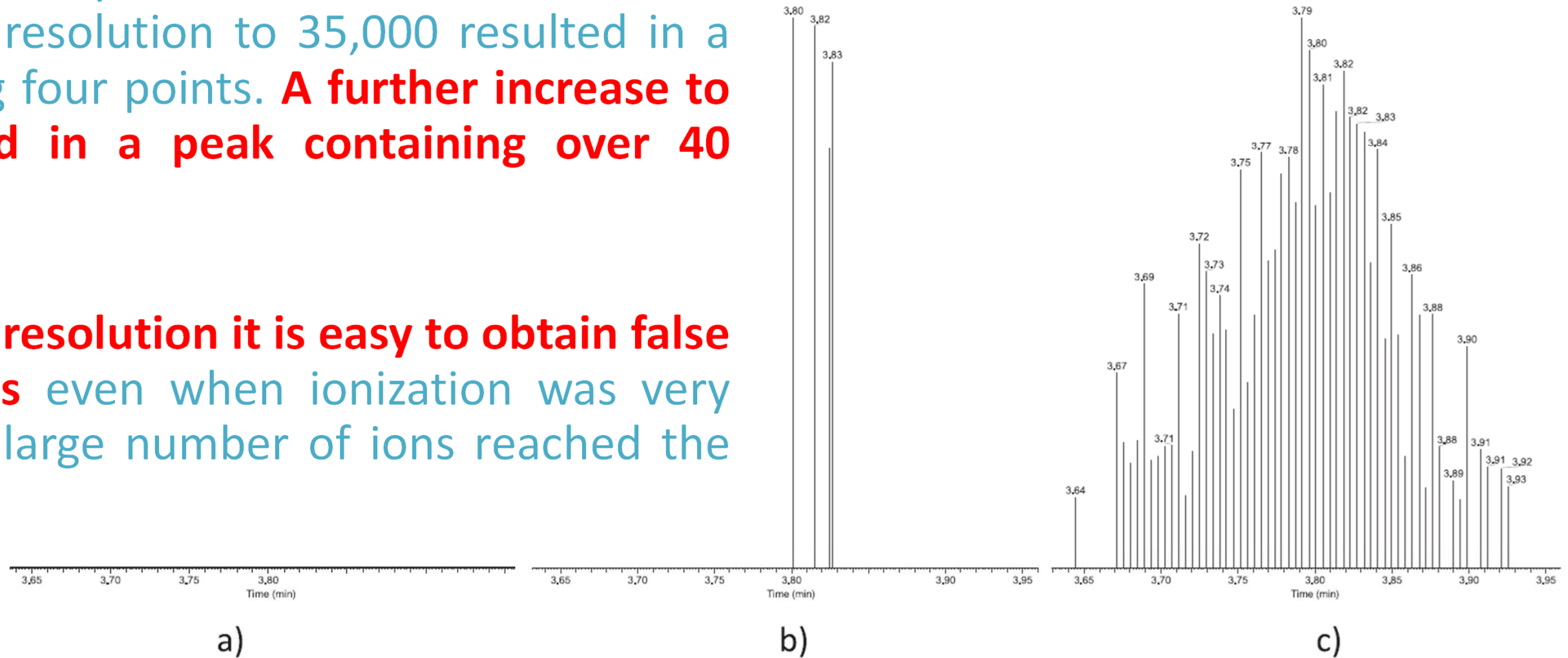
- Avoiding false positives and negatives
- Allowing a retrospective evaluation



# Target screening of 170 pesticides

“pymetrozine in orange at a 10 g/kg level **was not detected** when analysed **with a 17,500 resolution**. An increase of resolution to 35,000 resulted in a peak containing four points. **A further increase to 70,000 resulted in a peak containing over 40 points**”

“**With a 17,500 resolution it is easy to obtain false negative results** even when ionization was very efficient and a large number of ions reached the detector.”



**Fig. 4.** Increase of number of points per peak at higher resolution. 10 µg/kg of pymetrozine in orange extract, peak obtained with resolution (a) 17,500, (b) 35,000 and (c) 70,000.

# Suspect screening

Known products WITHOUT standards available  
or  
Suspected unknown pollutants

Use of prediction system for transformation products:

- University of Minnesota Pathway Prediction System (UM-PPS: <http://eawag-bbd.ethz.ch/>)
- CATABOL (<http://oasis-lmc.org/products/models/environmental-fate-and-ecotoxicity/catabol-301c.aspx>)
- PathPred (<http://www.genome.jp/tools/pathpred/>)
- Meteor (<http://www.lhasalimited.org/products/meteor-nexus.htm>)

Fragmentation pathway (should be similar to pollutant)



# Suspect screening of OH-PAH

LC-Orbitrap HESI negative, resolution 120,000 (FWHM):

- Strategy : Targeted and suspects

Sampling of air particulates

- glass fiber filters.
- Extraction: ASE (accelerated solvent extraction system)
- Concentration: solid phase extraction (SPE).

20 hydroxylated polycyclic aromatic hydrocarbons detected:

- 9 using the target screening
- 11 tentatively identified and semi-quantified by suspect screening



# Non target screening

Unknown pollutants and unpredictable

FullScan HR-MS/MS, mass error < 5ppm

Post-acquisition data-processing	MZmine	<a href="http://mzmine.sourceforge.net/">http://mzmine.sourceforge.net/</a>
	XCMS	<a href="https://xcmsonline.scripps.edu">https://xcmsonline.scripps.edu</a>
	EnviMass	<a href="http://www.eawag.ch/forschung/uchem/software/enviMass1">http://www.eawag.ch/forschung/uchem/software/enviMass1</a>
Comparison to database	MassBank	<a href="http://massbank.ufz.de/MassBank/">http://massbank.ufz.de/MassBank/</a>
	MetLin	<a href="http://metlin.scripps.edu/index.php">http://metlin.scripps.edu/index.php</a>
	m/zCloud	<a href="https://www.mzcloud.org/">https://www.mzcloud.org/</a>
Fragmentation rules	MOLGEN-MS	<a href="http://www.molgen.de/">http://www.molgen.de/</a>
	ACD/MS Fragmenter	<a href="http://www.acdlabs.com/products/adh/ms/ms_frag">www.acdlabs.com/products/adh/ms/ms_frag</a>
	MassFrontier	<a href="http://www.highchem.com/index.php/component/content/article?id=81">http://www.highchem.com/index.php/component/content/article?id=81</a>
	SIRIUS	<a href="http://bio.informatik.uni-jena.de/sirius2/">http://bio.informatik.uni-jena.de/sirius2/</a>

# Non target screening of pesticides metabolites

LC-Orbitrap HESI positive, Resolution 50,000 (FWHM)

- Strategy: suspect and non-target screening
- database: 240 pesticides metabolites
- « fragmentation-degradation »: non-target analysis

Sampling of air particulates (Valencia region)

- Quartz fiber filters
- Extraction by MAE
- Concentration, filtration

Identified pesticides and metabolites:

- 34 metabolites identified, 11 confirmed by standards
- 2 elucidated for non target screening

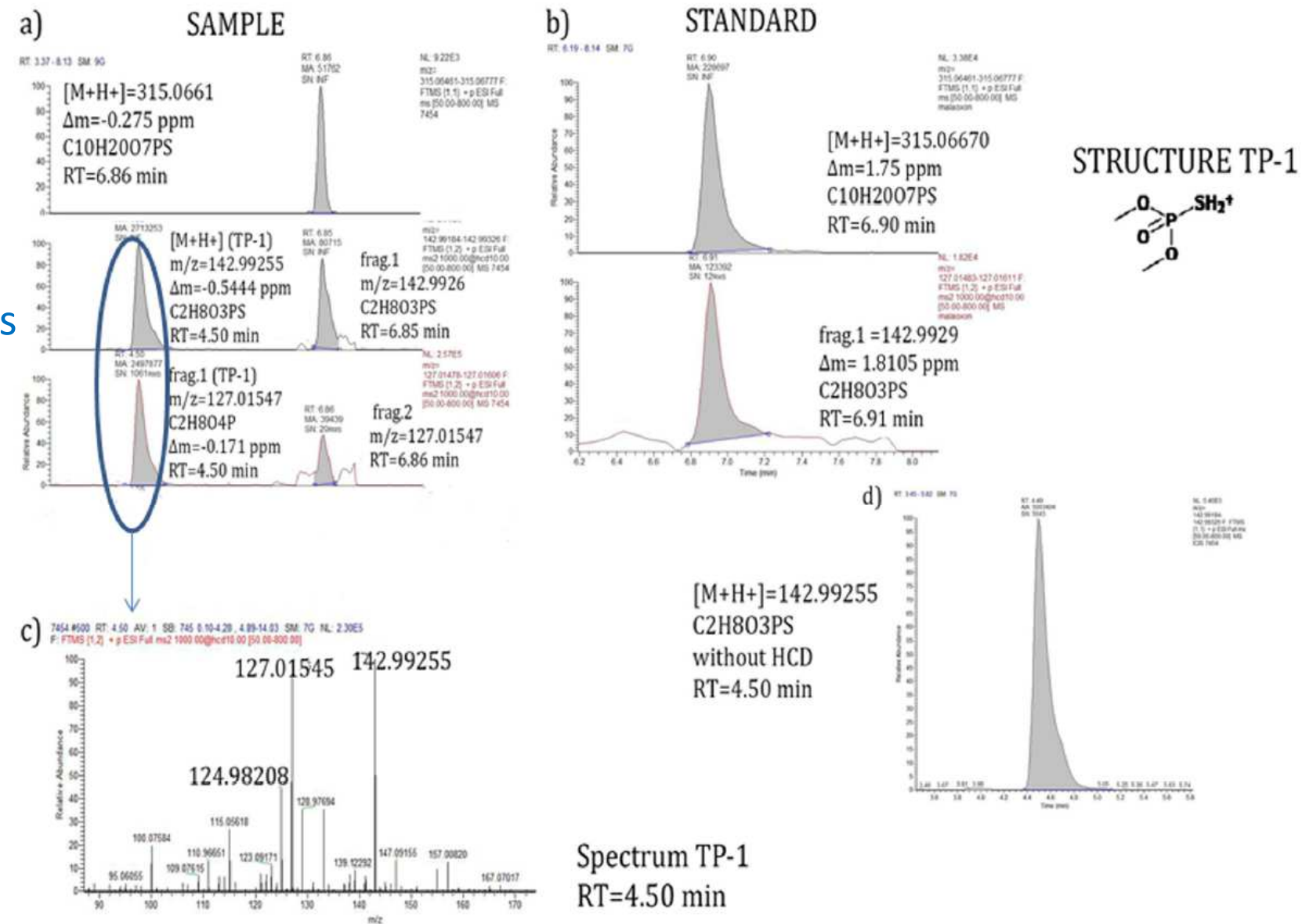
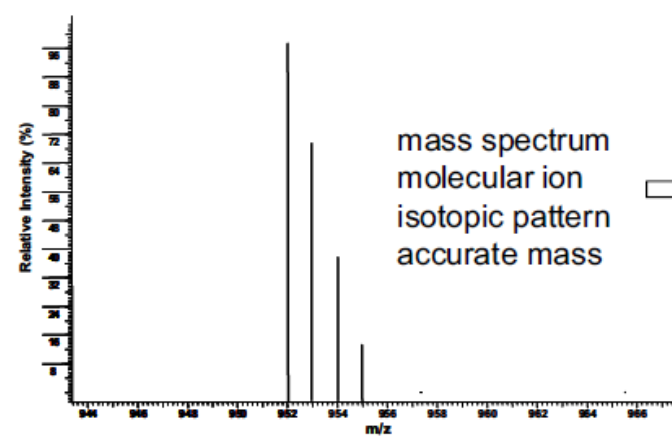


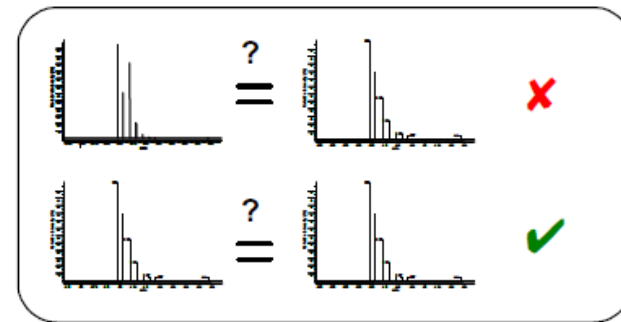
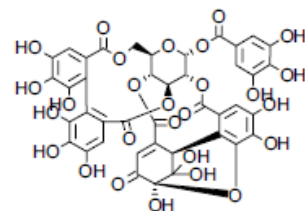
Fig. 4. (a) Extracted ion chromatograms (XIC) of malaoxon ( $m/z = 315.0661$ ), its fragments ( $m/z 142.9926$ ;  $m/z 127.01547$ ); and of a transformation product (TP-1) of malathion/malaoxon ( $m/z 142.9926$ ) obtained from air sample; (b) XIC of malaoxon standard and a characteristic ion fragment; (c) Accurate mass spectrum at 4.50 min, which corresponds to malathion/malaoxon transformation product-1 (TP-1); d) XIC of TP-1 ( $m/z = 142.99255$ ) without HCD.

# Mass accuracy and isotopic abundance accuracy



Formula Generator

No.	Formula	Mass
861	C61H22N4OP2S2	952.071
862	C61H22N4O3P2S	952.089
863	C61H23N4OP3S	952.081
864	C61H23N4O3P3	952.098
865	C61H24N4OP4	952.09
866	C61H29O4P3S3	952.097
867	C61H30O2P2S3	952.088
868	C61H31P3S3	952.08
869	C61H31O2P3S2	952.098
870	C61H32P4S2	952.09
871	C61H119N6O	951.945
872	C61H123O4S	951.914
873	C61H123O6	951.932
874	C61H124O2PS	951.906
875	C61H124O4P	951.924
876	C61H125O2P2	951.915
877	C61H126P3	951.907
878	C61H127N2S2	951.944
879	C61H127N2O2S	951.962



{exhaustive drill down}

Automatic Isotopic Pattern Filter



No.	Formula	Mass
1	C41H28O27	952.082
2	C41H28N8O20	952.142
3	C41H28N16O13	952.202
4	C41H28N24O6	952.262

DB Search

{fast approach but non comprehensive}

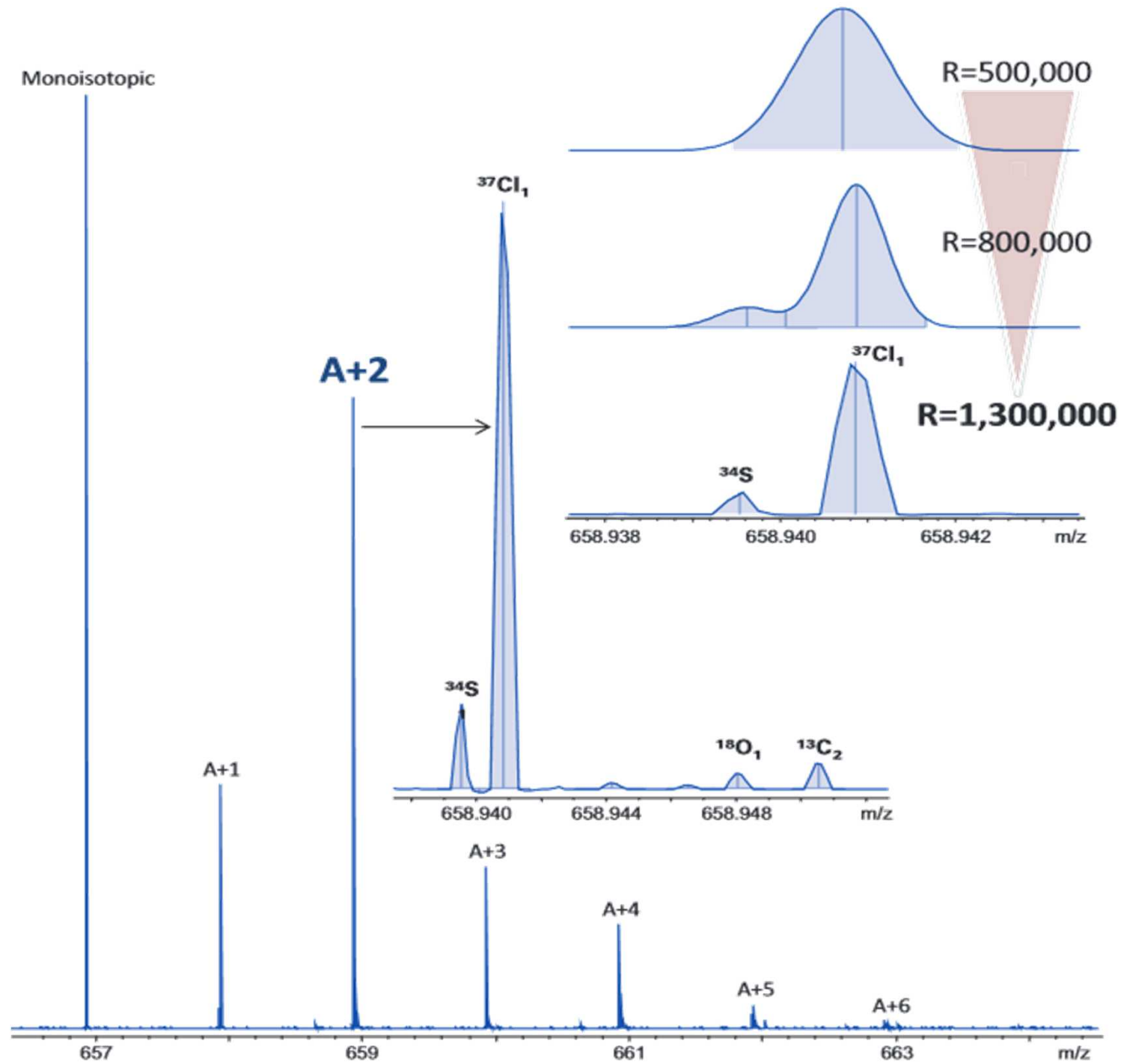
molecular isomer generator

{slow approach, needs constraints, comprehensive}

# Mass accuracy and isotopic abundance accuracy

Molecular mass (Da)	Without isotope abundance information					2% isotopic abundance accuracy	5% isotopic abundance accuracy
	10 ppm	5 ppm	3 ppm	1 ppm	0,1 ppm	3 ppm	5 ppm
150	2	1	1	1	1	1	1
200	3	2	2	1	1	1	1
300	24	11	7	2	1	1	6
400	78	37	33	7	1	2	13
500	266	115	64	21	2	3	33
600	505	257	155	50	5	4	36

# Ultra high resolution



Isotopic Fine Structure 09-2016 (1847374), Bruker



# Conclusions

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## DI-FTMS

Sample fingerprint

Fast ( $< 5$  min)

Less solvent

Isomers separation

Ion suppression

Dynamic range

Pollutant in low concentration



# Conclusions

## LC-FTMS

- Isomers separation
- Increase of dynamic range
- Separation of mixture
- Decrease of matrix effect on FT-MS
- Better selectivity

- Increase analysis time ( 10 to 50 min)
- Matrix effect on retention time
- Optimisation compounds dependent
- Solvent



# Conclusions

LC-FTMS vs LC-QqQ

Avoiding false negatives and false positives

Retrospective analysis

Resolution 500,000 to 10,000,000

Mass accuracy < 0,1 ppm

Femtogram-level sensitivity

Fast scanning rate at 15 Hz

Development of new data processing algorithms,  
software and databases



# Outlook

- Improve scan rate: better chromatographic resolution
- Processing software improvement from raw data to biological interpretation
- Hyphenated techniques and multi-dimensional chromatography
- Standardizing data and metadata standards, bioinformatics tools



Thank you!

